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**Evaluation of granular anaerobic ammonium oxidation process for
the disposal of pre-treated swine manure**

Shou-Qing Ni*, Ning Yang

*Shandong Provincial Key Laboratory of Water Pollution Control and Resource Reuse, School of
Environmental Science and Engineering, Shandong University, Jinan 250100, China*

* Corresponding author. Tel.: +86 531 8836 5660; fax: +86 531 8836 4513.
E-mail address: sqni@sdu.edu.cn.

12 **Abstract:** With rising environmental concerns on potable watersafety and eutrophication,
13 increased media attention and tighter environmental regulations, managing animal wastes in an
14 environmentally responsible and economically feasible way can be a challenge. In this study, the
15 possibility of using granular anammox process for ammonia removal from swine waste treatment
16 water was investigated. A rapid decrease of NO_2^- -N and NH_4^+ -N was observed during incubation
17 with wastewater from an activated sludge deodorization reactor and anaerobic digestion-partial
18 oxidation treatment process treating swine manure and its corresponding control artificial
19 wastewaters. Ammonium removal dropped from $98.0 \pm 0.6\%$ to $66.9 \pm 2.7\%$ and nearly absent
20 when the organic load in the feeding increased from 232 mg COD/L to 1160 mg COD/L and
21 2320 mg COD/L. The presence of organic carbon had limited effect on nitrite and total nitrogen
22 removal. At a COD to N ratio of 0.9, COD inhibitory organic load threshold concentration was
23 727 mg COD/L. Mass balance indicated that denitrifiers played an important role in nitrite,
24 nitrate and organic carbon removal. These results demonstrated that anammox system had the
25 potential to effectively treat swine manure that can achieve high nitrogen standards at reduced
26 costs.

27 **Key words:** Anammox; granular sludge; nitrogen; organic matter; swine manure; mass balance

28

29 **1. Introduction**

30 Large concentrated swine feeding operations throughout the world are presently producing a
31 huge amount of manure with abundant nitrogen and phosphorus as well as organic matter (Zhang
32 et al., 2006). Liquid swine manure can provide essential nutrients for plant growth. On the other
33 hand, continuing land application for manure disposal could result in excessive nutrient loss from
34 soil to water, causing eutrophication that deteriorates water quality (Karlen, Cambardella &
35 Kanwar, 2004). Manure also contributes to the production of greenhouse gas emissions (Thorman

36 [et al., 2007](#)). Usually, effluent from anaerobic wastewater treatment processes is characterized by
37 a high concentration of nitrogen and a low concentration of organic matters (i.e., a low C/N ratio)
38 ([Kataoka et al., 2002](#)). Biological nitrogen removal is achieved mostly by complete oxidation to
39 NO_3^- with surplus oxygen and subsequent reduction of NO_3^- to N_2 gas under anoxic conditions at
40 the expense of COD. If the C/N ratio in wastewater is low, additional carbon for denitrification is
41 required. Special attention needs to be given to N_2O gas emissions during biological nitrogen
42 removal process ([Hu et al., 2013](#); [Kong et al., 2013](#)). Therefore, there is an urgent call for
43 development of sustainable technologies for removals of N from swine manure with respect to
44 environmental and agricultural benefits.

45 Anaerobic ammonium oxidation (anammox) process is a novel, autotrophic and cost-effective
46 alternative to the traditional biological nitrogen removal process ([Ni & Zhang, 2013](#)). The
47 discovery of anammox process brought revolutionary changes to conventional biological nitrogen
48 removal from wastewater. Some unique characteristics make anammox process to be a promising
49 and sustainable technique ([Abma et al., 2007](#)), such as low biomass yield, no need for aeration
50 and no addition of external carbon sources ([Ni et al., 2010a](#)). In comparison to traditional
51 nitrification-denitrification process, this autotrophic process consumes 100% less biodegradable
52 organic carbon and at least 50% less oxygen ([Tal, Watts & Schreier, 2006](#)).

53 A long start-up period is expected in anammox process due to the slow growth rate of
54 anammox bacteria ([Strous et al., 1998](#)). Shortening anammox process start-up period by reducing
55 wash-out potential of anammox biomass becomes an important strategy for full-scale application.
56 Different types of reactor design have been used to minimize the wash-out of anammox biomass
57 including continuous stirred-tank reactor, anaerobic biological filtrated reactor, sequencing batch
58 reactor (SBR), up-flow reactor and biofilm reactor ([Imajo, Tokutomi & Furukawa, 2004](#); [Isaka,](#)
59 [Sumino & Tsuneda, 2007](#); [Strous et al., 1998](#); [van Dongen, Jetten & van Loosdrecht, 2001](#)).

60 Faster growth of anammox bacteria was achieved in a membrane bioreactor (the doubling time
61 was less than 10 days), resulting in an unprecedented purity of the enrichment of 97.6% ([van der
62 Star et al., 2008](#)). The formation of compact aggregates was reported to maintain a large amount
63 of active anammox biomass in a reactor ([Imajo, Tokutomi & Furukawa, 2004](#)). Therefore,
64 granulation is an alternative approach for anammox enrichment.

65 Only a few studies have investigated the possibility of using the anammox process for
66 ammonia removal from swine waste treatment water ([Ahn, Hwang & Min, 2004](#); [Hwang et al.,
67 2005](#); [Molinuevo et al., 2009](#); [Waki et al., 2007](#)). However, there is still a big gap regarding the
68 performance of anammox granules for the treatment of swine manure. The objective of this study
69 was to develop a potential swine manure treatment process that can achieve high nitrogen
70 standards at reduced costs by investigating the performance of anammox granular process fed
71 with pre-treated swine manure effluent.

72 **2. Materials & methods**

73 *2.1. Granules cultivation and reactor operation*

74 Two lab-scale up-flow anaerobic sludge blanket (UASB) reactors were inoculated with 900 mL
75 anammox granules from a running UASB reactor ([Ni et al., 2011](#)). The mixed liquor suspended
76 solid and mixed liquor volatile suspended solid of the seed sludge were 4.24 g/L and 3.35 g/L,
77 respectively. The reactors were running in a continuous mode at a HRT of approximately 1.0
78 days. The effluent was recycled from the bottom of the reactor. One reactor was designated as the
79 control one.

80 The reactors were operated at 35 °C with a working volume of 3.0 L. Different sizes of gravel
81 were placed in the bottom of the reactors. The pH in the reactor was controlled approximately 7.5
82 using CO₂ purge and the anoxic condition was created via argon gas. Before feeding with swine
83 manure, the reactor was pumped with synthetic wastewater prepared by adding ammonium and

84 nitrite to a mineral medium in the required amounts in the form of $(\text{NH}_4)_2\text{SO}_4$ and NaNO_2 . The
85 composition of the mineral medium was (g/L): KHCO_3 0.5, KH_2PO_4 0.0272, $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ 0.3,
86 $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ 0.18 and 1 mL trace elements solutions I and II (Ni et al., 2010a). The synthetic
87 wastewater was deoxygenated by flushing with argon gas before feeding to the reactor.

88 The effluents from an activated sludge deodorization reactor and anaerobic digestion-partial
89 oxidation treatment (AD-PO) process treating swine manure were collected. The effluent from
90 the activated sludge deodorization reactor contained 220 mg/L NH_4^+ -N, 265 mg/L NO_2^- -N, 125
91 mg/L NO_3^- -N, and 230 mg/L COD. The effluent from the AD-PO process contained 610 mg/L
92 NH_4^+ -N, 650 mg/L NO_2^- -N, 1350 mg/L NO_3^- -N, and 2320 mg/L COD. Both reactors were
93 initially fed with synthetic wastewater for 35 days. Then one reactor was fed with the effluent
94 from the activated sludge deodorization reactor without dilution and the other one with the
95 effluent from the AD-PO process, which was done gradually in increments of 10%, 20%, 50%
96 and 100% (v/v).

97 2.2. EPSs extraction and analysis

98 The EPSs in the granules were extracted using cation exchange resin (CER). In general, sludge
99 samples were harvested by centrifugation at 3000 rpm for 15 min at 4 °C and then the sludge
100 pellets were re-suspended in phosphate buffer solution (pH 7.0) and the solution was transferred
101 to an extraction bottles, followed by the CER addition with a dosage of 75 g/g suspended solids.
102 These suspensions were stirred at 600 rpm at 4 °C for 2 hours. After removing the settled CER,
103 the solutions were centrifuged at 8000 rpm for 30 min to remove remaining sludge components.
104 The supernatants were then filtered through 0.45 μm cellulose membranes and used as the EPSs
105 fraction for protein and carbohydrate analyses. The protein content in the EPSs was determined
106 according to the Bradford protein assay with bovine serum albumin as the standard (Bradford,

107 1976). The carbohydrate content in the EPSs was measured using the Anthrone method with
108 glucose as the standard (Gaudy, 1962). The total EPSs content was measured as the sum of these
109 two substances.

110 2.3. DNA extraction and quantitative real-time polymerase chain reaction (PCR)

111 Total genomic DNA was extracted by the modified 2% cetyl trimethyl ammonium bromide-
112 based protocol (Allen et al., 2006). Genomic DNA preparation was determined with an ND-1000
113 NanoDrop spectrophotometer (NanoDrop Technologies, Wilmington, DE, USA) and purified
114 DNA samples were stored in sterile deionized water at -20 °C until used. Quantitative PCR was
115 then processed based on the description of literature (Ni et al., 2011).

116 2.4. Analysis

117 Ammonia was measured by selective electrode according to the Standard Methods (APHA,
118 AWWA & WEF, 1998). Nitrite and nitrate concentrations were determined by ion-
119 chromatography (DX 500, Dionex, USA). The measurement of COD was carried out according
120 to the Standard Methods 5220 (APHA, AWWA & WEF, 1998). The SS and VSS were
121 determined by the weighing method after being dried at 103-105 °C and burnt to ash at 550 °C
122 (APHA, AWWA & WEF, 1998). For the electron microscopy observation, samples were fixed
123 with 2% paraformaldehyde and 2% glutaraldehyde in 0.1 M cacodylate buffer at 4 °C for 24 h.
124 Samples were then prepared following the method of Ni et al. (2011). For the transmission
125 electron microscopy (TEM), images were captured using a JEM 2100 200kV scanning and
126 transmission electron microscope (Japan Electron Optic Laboratories, Peabody, MA). For the
127 scanning electron microscopy (SEM), morphology characteristics of the biomass specimens were
128 observed using a JEOL 5800LV SEM (JEOL, Peabody, MA).

129 3. Results and Discussion

130 3.1. Control reactor performance and characteristics of anammox granules

131 Feeding with synthetic wastewater, the control experiment was carried out at a HRT of 1.0
132 days and the influent NH_4^+ -N to NO_2^- -N ratio was kept at around 1.0. Stable performance was
133 realized in several days after the addition of anammox granules. The reactor was run for 35 days
134 with high substrate removal. The average effluent ammonia and nitrite concentrations were $1.0 \pm$
135 0.4 and 0.6 ± 0.9 mg N/L, respectively (Figure 1), leading to the ammonia and nitrite removal
136 efficiencies of $98.0 \pm 0.8\%$ and $98.9 \pm 1.7\%$. Due to the production of nitrate by anammox
137 process, the total nitrogen (TN) removal efficiency was only $83.6 \pm 1.1\%$.

138 During the experiment, the granules were sampled for the microscope observation. As shown
139 in Figure 2A, the granules in the reactor were reddish, semitransparent and easy to congregate
140 with each other. Each part of the granules was densely incorporated with others, which favored
141 the granules joining tightly and existing stably. This structure was possibly formed due to the
142 shear forces of the effluent recirculation currents (Ni et al., 2011). Spherical shaped bacteria,
143 which were supposed to be anammox bacteria (Jetten et al., 1999), were observed (Figure 2B).
144 Transmission electron micrograph shows that anammox bacterial cells have an irregular
145 morphology (Figure 2C). In this paper, the cells displayed an identical pattern of organization to
146 other anammox species (Kartal et al., 2008).

147 From the SEM image (Figure 2B), anammox cells were surrounded by bacterial extracellular
148 polymeric substances (EPS). EPS were believed to play a fundamental role during the formation
149 of anammox granules (Ni et al., 2010b). Generally, bacterial EPS, consisting of polysaccharides,
150 proteins, nucleic acids, and lipids, are sticky materials secreted by microorganisms, acting as
151 cementing substances in biofilms and flocs (Characklis & Marshall, 1990; Frolund et al. 1996).
152 Proteins and carbohydrates were reported to be the dominant components in the extracted EPS
153 and therefore were usually employed to represent the EPS content. During the experiment, the
154 proteins and carbohydrates contents in the extracted EPSs of the granules were analyzed. The

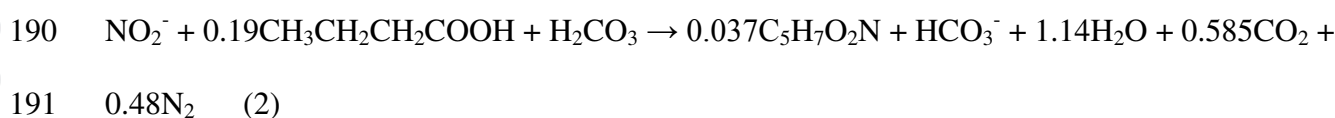
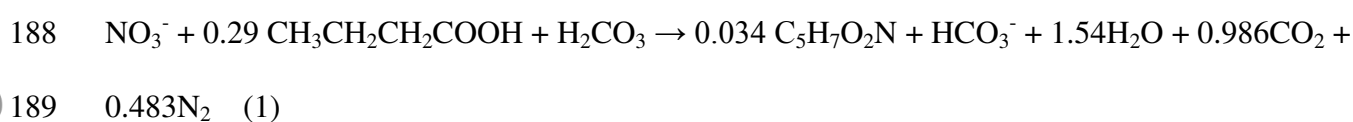
155 total EPSs content was measured as the sum of these two substances. The proteins and
156 carbohydrates in anammox granules were 56.7 ± 2.8 and 65.7 ± 3.2 mg/g VSS with a
157 protein/carbohydrate (PN/PS) ratio of approximately 0.9. The total EPSs contents in anaerobic
158 and aerobic granules were around 60 mg/g VSS (Wu et al., 2009; Zheng & Yu, 2007),
159 substantially lower than that for the anammox granules (total EPSs content was about 122.4 mg /
160 g VSS) in this study. The PN/PS ratios were higher than 2.0 for anaerobic, aerobic and nitrifying
161 granules (Martinez et al., 2004; Wu et al., 2009; Zheng & Yu, 2007), while it was lower than 1.0
162 for denitrifying granules (Bhatti et al., 2001), similar to that of this study. This suggested that
163 proteins might be the key EPS constituents for anaerobic, aerobic, and nitrifying granules, but
164 carbohydrates might play a significant function in the development of denitrifying and anammox
165 granules.

166 Quantitative real-time PCR analysis was used to quantify the microbial community of the
167 granules in the reactor, using the assay based on the 16S rRNA gene-specific set of primers
168 AMX809F/AMX1066R. The data indicated that anammox bacteria comprised about 91% cells in
169 the microorganisms' community, resulting in high NH_4^+ -N and NO_2^- -N removal efficiencies.
170 FISH images also showed that anammox bacteria constituted the majority of cells in the
171 community.

172 3.2. Nitrogen removal from pretreated swine manure

173 After more than a month stable operation, ammonium and nitrite removal rates in both reactors
174 reached over 95%, demonstrating that both anammox granular reactors were ready for further
175 study. Reactor I was fed with the effluent, which contained 220 mg/L NH_4^+ -N, 265 mg/L NO_2^- -N,
176 125 mg/L NO_3^- -N and 230 mg/L COD, from the activated sludge deodorization reactor for about
177 50 days. As shown in Figure 3, during the late 22 days, NH_4^+ -N, NO_2^- -N, and TN removal rate
178 were $92.2 \pm 1.5\%$, $99.3 \pm 0.9\%$, and $72.0 \pm 1.4\%$, respectively. Ammonium and nitrite removal

179 rates were very high, indicating the good activities of anammox microorganisms. Due to the
180 existence of NO_3^- -N in the feeding, the calculated TN removal rate ($[\text{removed NH}_4^+\text{-N} + \text{NO}_2^-\text{-N}$
181 $+ \text{NO}_3^-\text{-N}]/[\text{influent NH}_4^+\text{-N} + \text{NO}_2^-\text{-N} + \text{NO}_3^-\text{-N}]$) was only 72.0%. Besides anammox bacteria,
182 other species such as denitrifiers may contribute to nitrogen removal from wastewater. Process
183 stoichiometry was calculated to get a deep insight of their relations (Figure 3). The stoichiometry
184 molar ratios of $\text{NO}_2^-\text{-N}$ to $\text{NH}_4^+\text{-N}$ conversion and $\text{NO}_3^-\text{-N}$ removal to $\text{NH}_4^+\text{-N}$ conversion were
185 1.30 ± 0.01 and 0.14 ± 0.008 . More nitrite was removed and fewer nitrates were produced. This
186 finding indicated that organic matters enhanced the nitrogen removal by favoring the denitrifiers
187 and they consumed the surplus nitrite and produced nitrate (Eq. (1) and (2)) (Ni et al., 2012).



192 Reactor II was fed with the effluent, which contained 610 mg/L $\text{NH}_4^+\text{-N}$, 650 mg/L $\text{NO}_2^-\text{-N}$,
193 1350 mg/L $\text{NO}_3^-\text{-N}$, and 2320 mg/L COD, from the AD-PO process for about 2 months. The
194 presence of organic matters was found to affect anammox process adversely (van de Graaf et al.,
195 1996). Anammox microorganisms could not compete with denitrifiers for nitrite and may result
196 in complete inactivation of anammox activity under high organic matter concentration (Güven et
197 al., 2005; Molinuevo et al., 2009). So the feeding in reactor II was done gradually in increments
198 of 10%, 20%, 50% and 100% (Figure 4). The addition of 10% of AD-PO effluent (organic load
199 of 232 mg COD/L) resulted in up to $98.0 \pm 0.6\%$ of high ammonium removal, compared with
200 ammonium removal for activated sludge deodorization reactor effluent ($92.2 \pm 1.5\%$, organic
201 load of 230 mg COD/L). The difference was caused by higher ammonium concentration (~ 220
202 mg N/L) of activated sludge deodorization reactor effluent than that (~ 60 mg N/L) after AD-PO

203 treatment. As high contents of free ammonia were toxic to anammox process (Waki et al., 2007),
204 pretreatments, for example partial oxidation of ammonia to oxidized nitrogen, may facilitate
205 anammox reaction. Then, the feeding rate was increased gradually to organic loads of 464, 1160,
206 and 2320 mg COD/L, corresponding to 20%, 50% and 100% of AD-PO effluent. Ammonium
207 removal rates were $88.0 \pm 1.0\%$ and $66.9 \pm 2.7\%$ when 464 and 1160 mg COD/L were pumped
208 into the reactor. When 2320 mg COD/L was added, the ammonium removal was dropped quickly
209 to nearly absent.

210 However, nitrite removal was seldom affected by organic loading rate. Most time, over 95% of
211 nitrite removal was achieved (Figure 4). The calculated TN removal rate was less than 50% due
212 to high concentration of NO_3^- -N in the feeding. The stoichiometry molar ratios of NO_2^- -N to
213 NH_4^+ -N conversion were 1.07 ± 0.01 and 1.18 ± 0.01 when 232 and 464 mg COD/L were applied,
214 close to the theoretical value (Strous et al., 1998). And this value increased to 1.58 ± 0.07 at
215 1160 mg COD/L. When organic load of 2320 mg COD/L was achieved, ratios of 3.82-8.39 were
216 obtained and as time went by, this ratio increased up to 44.0. In this case the heterotrophic
217 denitrification was the major reaction involved in ammonium removal (Molinuevo et al., 2009).
218 Results from the mass balance showed that the participation of anammox process in the total
219 ammonium and nitrite removal decreased when high percent of AD-PO effluent was
220 implemented, which was replaced by the denitrification part.

221 The physiological changes of biomass were also observed. When organic load of 2320 mg
222 COD/L was achieved, the disintegration of biomass was registered. The red granules began to
223 turn black and more aggregated biomass disassembled to small parts. Due to the change of
224 running conditions by the addition of more organic matters, anammox communities decreased
225 and denitrifiers took charge of ammonium removal eventually. In this situation, slowly growing
226 anammox bacteria ($Y = 0.066 \pm 0.01$) are incapable of competing with denitrifiers with higher

227 growth yield ($Y = 0.3$). FISH images also showed that there was a reduction in the number of
228 anammox cells when 2320 mg COD/L was added in comparison with the abundance of anammox
229 microorganisms at organic load of 232 mg COD/L.

230 *3.3. Effect of organic matters on anammox performance*

231 Literature review showed that high content of organic matters usually inhibited anammox
232 activity. In this study, at a COD to N ratio of 0.9, COD inhibitory organic load threshold
233 concentration was 727 mg COD/L (Figure 5). Previously, we found the threshold were 308 mg
234 COD/L and 3.1 for COD to N ratio (Ni et al., 2012). Both organic matter concentrations and
235 COD to N ratios affect the performance of anammox bacteria without a general agreement
236 (Chamchoi, Nitorisvut & Schmidt, 2008; Molinuevo et al., 2009). Batch tests showed that 25 and
237 50 mM acetate resulted in 70 and 22% inhibition in anammox process (Dapena-Mora et al.,
238 2007). Güven et al (2005) indicated that even 0.5 mM of methanol resulted in the immediate and
239 complete inactivation of anammox activity. About 300 mg COD/L (COD to N ratio of 2) was
240 found to inactivate or eradicate anammox communities under concurrent operation of anammox
241 and denitrification (Chamchoi, Nitorisvut & Schmidt, 2008). At a COD to N ratio of 0.5, COD
242 inhibitory organic load threshold concentration (defined when ammonia removal dropped to 80%)
243 were 142 and 242 mg/L when treating different wastewaters (Molinuevo et al., 2009).

244 To further understand the effect of organic matters on anammox performance, mass balance
245 evaluation of participation of different processes was done as illustrated in Table 1. At low COD
246 to N ratios, variation of COD to N ratio had limited effect on anammox performance. At COD to
247 N ratio of 0.9, anammox accounted for 98.9% nitrite removal, while at COD to N ratio of 0.4,
248 anammox accounted for 88.5% nitrite removal (Table 1). The difference was mainly caused by
249 influent substrate concentration. Somehow, COD removal in reactor I had higher efficiency, i.e.
250 over 70%. Less than 50% COD was removed in reactor II at all conditions. COD removal was

251 resulted from denitrification by the denitrifying communities mainly using nitrite as electron
252 acceptor. Hence the competition for nitrite as electron acceptor between the denitrifying bacteria
253 and the anammox communities existed in the reactors. Nitrite (ammonia) consumption via
254 anammox was 98.9% (98.0%) when influent COD and ammonium concentrations were 232 mg/L
255 and 67 mg N/L, but it decreased sharply to 27.5% (<5%) at 2320 mg COD/L and 615 NH₄⁺-N/L
256 (Table 1), indicating that denitrification prevailed anammox process gradually. Denitrification
257 helped to remove nitrite and nitrate when organic matter was available and will become the
258 dominant route in the reactor in time.

259

260 **4. Conclusion**

261 Nitrogen removal has become a major focus in swine manure treatment since nitrogen is the
262 nutrient concerning the application amount of the manure produced in accordance with an
263 increasing number of governmental regulations. As a novel, autotrophic and cost-effective
264 alternative to the traditional biological nitrification/denitrification removal process, anammox
265 process was proved to be effective for swine manure nitrogen removal. With increasing organic
266 matters, ammonium removal via anammox decreased and the role of denitrifiers in nitrite, nitrate
267 and COD removal became significant, proved by mass balance. The introduction of organic
268 matters favored the growth of denitrifiers. At low COD to N ratios, variation of COD to N ratios
269 had limited effect on anammox performance.

270 **Conflict of Interests**

271 The authors declare that there is no conflict of interests.

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278 **References**

279 Abma WR, Schultz CE, Mulder JW, van der Star WR, Strous M, Tokutomi T, van Loosdrecht
280 MC. 2007. Full-scale granular sludge Anammox process. *Water Science and Technology*
281 55:27-33.

282 Ahn YH, Hwang IS, Min KS. 2004. ANAMMOX and partial denitrification in anaerobic nitrogen
283 removal from piggery waste. *Water Science and Technology* 49:145-153.

284 Allen GC, Flores-Vergara MA, Krasnyanski S, Kumar S, Thompson WF. 2006. A modified
285 protocol for rapid DNA isolation from plant tissues using cetyltrimethylammonium bromide.
286 *Nature Protocols* 1:2320-2325.

287 APHA, AWWA, WEF. 1998. *Standard methods for the examinations of water and wastewater,*
288 *20th edition.* Washington, DC: American Public Health Association.

289 Bhatti ZI, Sumida K, Rouse JD, Furukawa K. 2001. Characterization of denitrifying granular
290 sludge treating soft groundwater in an upflow sludge-blanket reactor. *Journal of Bioscience*
291 *and Bioengineering* 91:373-377.

292 Bradford MM. 1976. Rapid and sensitive method for the quantitation of microgram quantities of
293 protein utilizing the principle of protein-dye binding. *Analytical Biochemistry* 72:248-254.

294 Chamchoi N, Nitisornvut S, Schmidt JE. 2008. Inactivation of ANAMMOX communities under
295 concurrent operation of anaerobic ammonium oxidation (ANAMMOX) and denitrification.
296 *Bioresource Technology* 99:3331-3336.

297 Characklis WG, Marshall KC. 1990. Biofilms: a basis for an interdisciplinary approach. In:
298 Characklis WG, Marshall K C, eds. *Biofilms.* New York: Wiley-Interscience, 3-15.

- 299 Dapena-Mora A, Fernandez I, Campos JL, Mosquera-Corral A, Mendez R, Jetten MSM. 2007.
300 Evaluation of activity and inhibition effects on Anammox process by batch tests based on the
301 nitrogen gas production. *Enzyme and Microbial Technology* 40:859-865.
- 302 Frolund B, Palmgren R, Keiding K, Nielsen PH. 1996. Extraction of extracellular polymers from
303 activated sludge using a cation exchange resin. *Water Research* 30:1749-1758.
- 304 Gaudy AF. 1962. Colorimetric determination of protein and carbohydrate. *Industrial Water*
305 *Wastes* 7:17-22.
- 306 Güven D, Dapena A, Kartal B, Schmid MC, Maas B, van de Pas-Schoonen K, Sozen S, Mendez
307 R, Op den Camp HJ, Jetten MS, Strous M, Schmidt I. 2005. Propionate oxidation by and
308 methanol inhibition of anaerobic ammonium-oxidizing bacteria. *Applied and Environmental*
309 *Microbiology* 71:1066-1071.
- 310 Hu Z, Zhang J, Liang S, Xie H. 2013. Impact of carbon source on nitrous oxide emission from
311 anoxic/oxic biological nitrogen removal process and identification of its emission sources.
312 *Environmental Science and Pollution Research* 20:1059-1069.
- 313 Hwang IS, Min KS, Choi E, Yun Z. 2005. Nitrogen removal from piggery waste using the
314 combined SHARON and ANAMMOX process. *Water Science and Technology* 52:487-494.
- 315 Imajo U, Tokutomi T, Furukawa K. 2004. Granulation of Anammox microorganisms in up-flow
316 reactors. *Water Science and Technology* 49: 155-163,
- 317 Isaka K, Sumino T, Tsuneda S. 2007. High nitrogen removal performance at moderately low
318 temperature utilizing anaerobic ammonium oxidation reactions. *Journal of Bioscience and*
319 *Bioengineering* 103: 486-490.
- 320 Jetten MSM, Strous M, van de Pas-Schoonen KT, Schalk J, van Dongen UG, van de Graaf
321 AA, Logemann S, Muyzer G, van Loosdrecht MC, Kuenen JG. 1998. The anaerobic oxidation
322 of ammonium. *FEMS Microbiology Review* 22:421-437.

- 323 Karlen DL, Cambardella CA, Kanwar RS. 2004. Challenges of managing liquid swine manure.
324 *Applied Engineering in Agriculture* 20:693-699.
- 325 Kartal B, van Niftrik L, Rattray J, van de Vossenberg JL, Schmid MC, Sinnighe Damsté
326 J, Jetten MS, Strous M. 2008. Candidatus 'Brocadia fulgida': an autofluorescent anaerobic
327 ammonium oxidizing bacterium. *FEMS Microbiology Ecology* 63:46-55.
- 328 Kataoka N, Suzuki T, Ishida K, Yamada N, Kurata N, Katayose M, Honda K. 2002. Field test of
329 methane fermentation system for treating swine wastes. *Water Science and Technology*
330 45:103-112.
- 331 Kong Q, Liang S, Zhang J, Xie H, Miao M, Tian L. 2013. N₂O emission in a partial nitrification
332 system: dynamic emission characteristics and the ammonium-oxidizing bacteria community.
333 *Bioresource Technology* 127:400-406.
- 334 Martinez F, Lema J, Mendez R, Cuervo-Lopez F, Gomez J. 2004. Role of exopolymeric protein
335 on the settleability of nitrifying sludges. *Bioresource Technology* 94:43-48.
- 336 Molinuevo B, García MC, Karakashev D, Angelidaki I. 2009. Anammox for ammonia removal
337 from pig manure effluents: effect of organic matter content on process performance.
338 *Bioresource Technology* 100:2171-2175.
- 339 Ni SQ, Fessehaie A, Lee PH, Gao BY, Xu X, Sung S. 2010b. Interaction of anammox bacteria
340 and inactive methanogenic granules under high nitrogen selective pressure. *Bioresource*
341 *Technology* 101:6910-6915.
- 342 Ni SQ, Gao BY, Wang CC, Lin JG, Sung S. 2011. Fast start-up, performance and microbial
343 community in a pilot-scale anammox reactor seeded with exotic mature granules. *Bioresource*
344 *Technology* 102:2448-2454.
- 345 Ni SQ, Lee PH, Fessehaie A, Gao BY, Sung S. 2010a. Enrichment and biofilm formation of
346 Anammox bacteria in a non-woven membrane reactor. *Bioresource Technology* 101:1792-1799.

- 347 Ni SQ, Ni J, Hu D, Sung S. 2012. Effect of organic matter on the performance of granular
348 anammox process. *Bioresource Technology* 110:701-705.
- 349 Ni SQ, Zhang J. 2013. Anaerobic Ammonium Oxidation: From Laboratory to Full-Scale
350 Application. *BioMed Research International* 2013, Article ID 469360.
- 351 Strous M, Heijnen JJ, Kuenen JG, Jetten MSM. 1998. The sequencing batch reactor as a powerful
352 tool for the study of slowly growing anaerobic ammonium-oxidizing microorganisms.
353 *Applied Microbiology and Biotechnology* 50:589-596.
- 354 Tal JEM, Watts J, Schreier HJ. 2006. Anaerobic ammonium-oxidizing (Anammox) bacteria and
355 associated activity in fixed-film biofilters of a marine recirculating aquaculture system.
356 *Applied and Environmental Microbiology* 72: 2896-2904.
- 357 Thorman RE, Chadwick DR, Harrison R, Boyels LO, Matthews R. 2007. The effect on N₂O
358 emissions of storage conditions and rapid incorporation of pig and cattle farmyard manure into
359 tillage land. *Biosystems Engineering* 97:501-511.
- 360 Van de Graaf AA, de Bruijn P, Robertson LA, Jetten MSM, Kuenen JG. Autotrophic growth of
361 anaerobic ammonium-oxidizing micro-organisms in a fluidized bed reactor. *Microbiology*
362 142:2187-2196.
- 363 Van der Star WR, Miclea AI, van Dongen UG, Muyzer G, Picioreanu C, van Loosdrecht MC.
364 2008. The membrane bioreactor: a novel tool to grow anammox bacteria as free cells.
365 *Biotechnology and Bioengineering* 101:286-294.
- 366 Van Dongen U, Jetten MSM, van Loosdrecht MCM. 2001. The SHARON®-Anammox® process
367 for treatment of ammonium rich wastewater. *Water Science and Technology* 44:153-160.
- 368 Waki M, Tokutomi T, Yokoyama H, Tanaka Y. 2007. Nitrogen removal from animal waste
369 treatment water by anammox enrichment. *Bioresource Technology* 98:2775-2780.

- 370 Wu J, Zhou HM, Li HZ, Zhang PC, Jiang J. 2009. Impacts of hydrodynamic shear force on
371 nucleation of flocculent sludge in anaerobic reactor. *Water Research* 43:3029-3036.
- 372 Zhang Z, Zhu J, King J, Li W. 2006. A two-step fed SBR for treating swine manure. *Process*
373 *Biochemistry* 41:892-900.
- 374 Zheng YM, Yu HQ. 2007. Determination of the pore size distribution and porosity of aerobic
375 granules using size-exclusion chromatography. *Water Research* 41:pp. 39-46.

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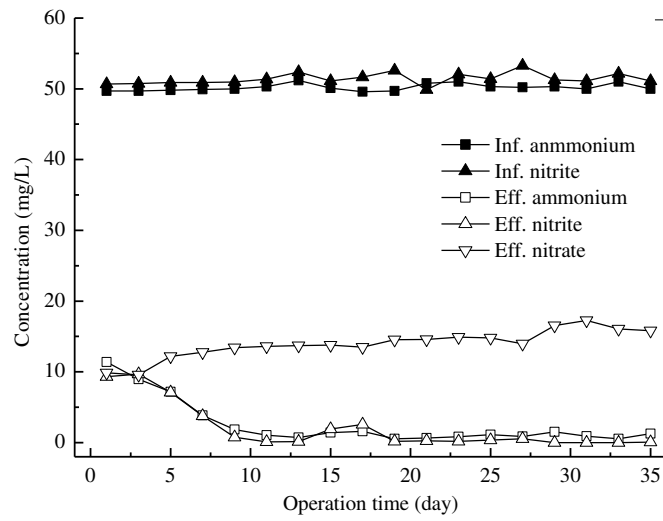
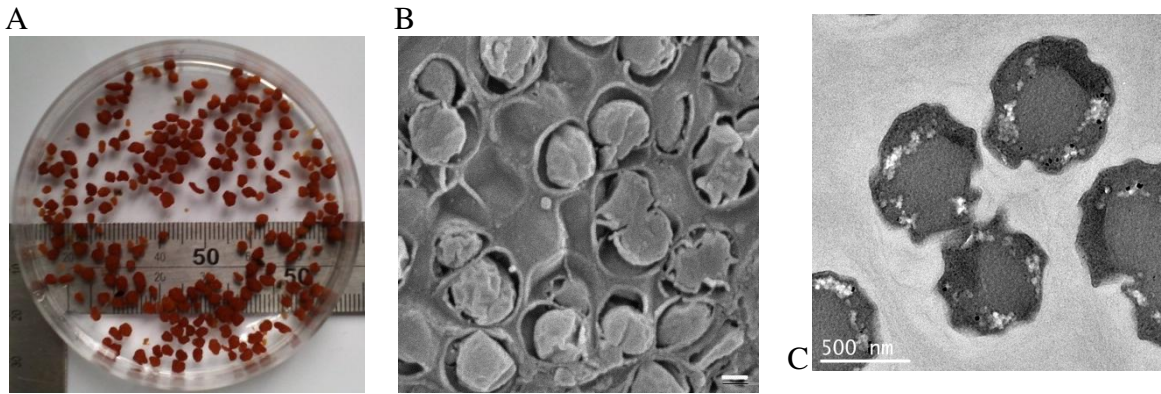


Figure 1: Nitrogen removal performance of control reactor feeding with synthetic wastewater.



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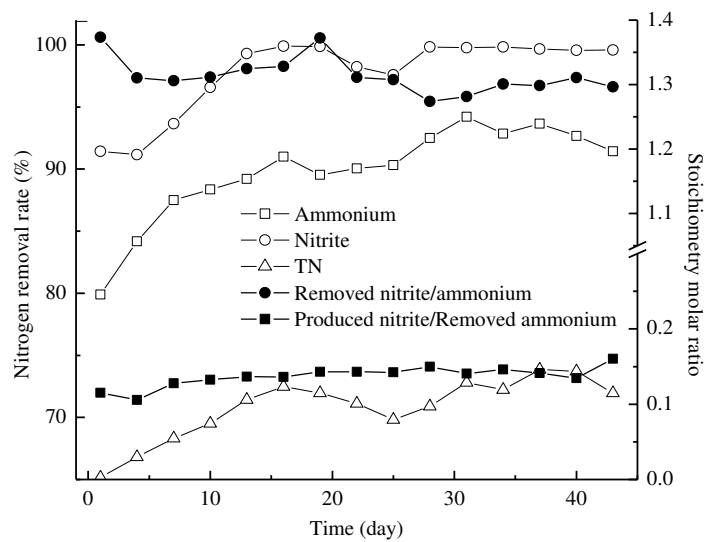
383 **Figure 2:** (A) Image showing the reddish anammox granules in a beaker. (B) Scanning electron

384 micrograph showing anammox bacteria surrounded by bacterial extracellular polymeric

385 substances (bar = 400 nm). (C) Transmission electron micrograph showing anammox bacterial

386 cells (bar = 500 nm).

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Figure 3: $\text{NH}_4^+\text{-N}$, $\text{NO}_2^-\text{-N}$ and TN removal (left axis) and process stoichiometry (right axis)

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during implementation of the effluent after activated sludge deodorization reactor.

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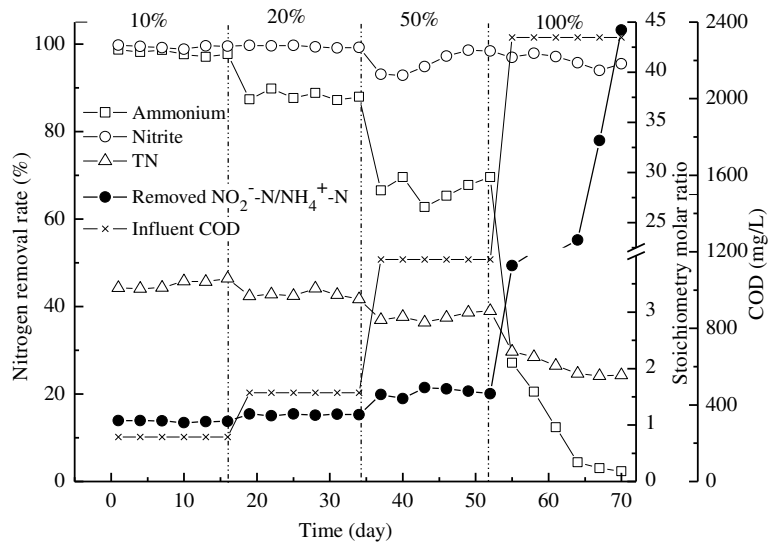


Figure 4: $\text{NH}_4^+ \text{-N}$, $\text{NO}_2^- \text{-N}$ and TN removal (left axis) and process stoichiometry (right axis)

during gradual implementation of the effluent anaerobic digestion-partial oxidation treatment.

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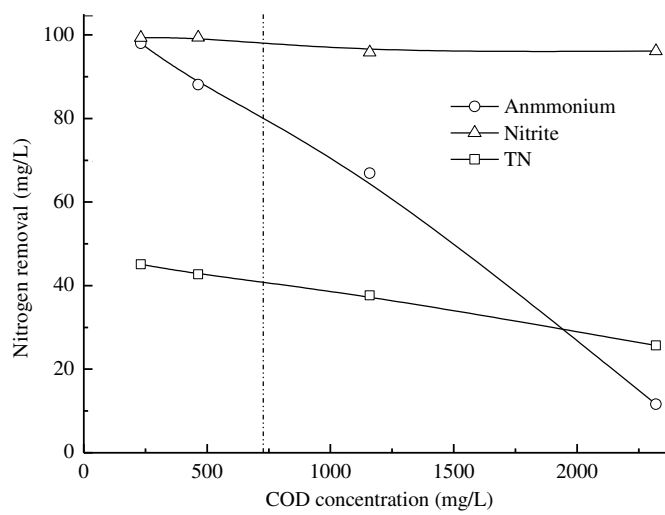


Figure 5: Effect of organic matter on anammox performance treating pretreated swine manure.

402 **Table 1:** Mass balance evaluation of participation of different processes.

COD conc. (mg/L)	COD to N ratio	NH ₄ ⁺ -N removal (mg N/L)	NO ₂ ⁻ -N removal (mg N/L)		NO ₃ ⁻ -N (mg N/L)			COD removal (mg/L)	Inf. NH ₄ ⁺ -N	NO ₂ ⁻ -N consumption via anammox (%)
			Anammox	Denitrification	Production ^a	Removal ^b	Final ^c			
0	-	48.8	51.0	-	15.8	-	15.8	-	50.0	100
232.0	0.9	65.5	68.8	0.8	21.0	11.8	9.2	41.8	67.0	98.9
464.0	0.9	115.3	121.1	15.4	37.4	15.7	21.7	86.6	121.1	88.7
1160.0	0.9	211.9	222.5	107.1	68.7	40.6	28.1	369.2	304.5	67.5
2320.0	0.9	166.9	175.2	462.6	54.1	29.0	25.1	1097.6	615.0	27.5
230.0	0.4	211.2	221.8	28.9	68.4	32.5	35.9	173.1	231.0	88.5

403 ^a Values denote nitrate produced via anammox process.

404 ^b Values means nitrate removed by denitrification process.

405 ^c Values stand for final concentration of effluent nitrate.