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Strategies for improved rhamnolipid production by *Pseudomonas aeruginosa* PA1

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Rhamnolipids are biosurfactants with potential for diversified industrial and environmental uses. The present study evaluated three strategies to increase the production of rhamnolipid-type biosurfactants produced by *Pseudomonas aeruginosa* strain PA1. The influence of pH, the addition of endogenous autoinducers and the use of a fed batch process were examined. The culture medium adjusted to pH 7.0 was the most productive. Furthermore, the pH of the culture medium had a measurable effect on the ratio of mono- and dirhamnolipids synthesized. At pH values below 7.3, the proportion of monorhamnolipids decreased from 45 to 24%. Additionally, recycling 20% of the spent culture medium where *P. aeruginosa* was grown up to the later stationary phase was responsible for a 100% increase in rhamnolipid volumetric productivity in the new culture medium. Finally, the use of fed batch operation under conditions of limited nitrogen resulted in a 3.8-fold increase in the amount of rhamnolipids produced (1.29 g L⁻¹ to 4.90 g L⁻¹, as rhamnose). These results offer promising paths to optimize processes for the production of rhamnolipids.

1 Strategies for improved rhamnolipid production by *Pseudomonas aeruginosa* PA1

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14

15 **Abstract**

16 Rhamnolipids are biosurfactants with potential for diversified industrial and environmental
17 uses. The present study evaluated three strategies to increase the production of rhamnolipid-type
18 biosurfactants produced by *Pseudomonas aeruginosa* strain PA1. The influence of pH, the
19 addition of endogenous autoinducers and the use of a fed batch process were examined. The
20 culture medium adjusted to pH 7.0 was the most productive. Furthermore, the pH of the culture
21 medium had a measurable effect on the ratio of mono- and dirhamnolipids synthesized. At pH
22 values below 7.3, the proportion of monorhamnolipids decreased from 45 to 24%. Additionally,
23 recycling 20% of the spent culture medium where *P. aeruginosa* was grown up to the later
24 stationary phase was responsible for a 100% increase in rhamnolipid volumetric productivity in
25 the new culture medium. Finally, the use of fed batch operation under conditions of limited
26 nitrogen resulted in a 3.8-fold increase in the amount of rhamnolipids produced (1.29 g L⁻¹ to
27 4.90 g L⁻¹, as rhamnose). These results offer promising paths to optimize processes for the
28 production of rhamnolipids.

29

30 **Keywords** – biosurfactant, rhamnolipid, homoserine lactones, *Pseudomonas aeruginosa*

31

32

33 Introduction

34 Rhamnolipids are biodegradable biological surfactants. They present low toxicity and high
35 resistance to extreme conditions of pH, salinity and temperature (Kesting *et al.*, 1996). Their
36 surface properties, such as emulsification, dispersion, foaming, detergency, wetting and
37 stabilization (Van Dyke *et al.*, 1993; Ishigami *et al.*, 1994; Desai and Banat, 1997; Bognolo,
38 1999), as well as their capacity for removing heavy metals (Torrens *et al.*, 1998; Lang and
39 Wullbrandt, 1999; Kitamoto *et al.*, 2002) and their anticorrosive capacity (Araujo *et al.*, 2013),
40 confer to these biosurfactants a variety of ecological (Rahman *et al.*, 2002) and commercial
41 applications in the oil, pharmaceutical, food and chemical industries (Sinumvayo and Ishimwe,
42 2015).

43 Rhamnolipids are produced mainly by *Pseudomonas aeruginosa*, a Gram-negative
44 bacterium that can be isolated from various habitats (water, soil or even plants). The control of
45 rhamnolipid production in *P. aeruginosa* is performed by a regulatory system called quorum
46 sensing that is controlled by acyl homoserine lactones. These lactones are signaling molecules
47 that, paired with the regulators LasR and RhIR, initiate the expression of the enzymes involved
48 in rhamnolipid synthesis (rhamnosyltransferases) (Ochsner *et al.*, 1994, 1995; Rahim *et al.*,
49 2001).

50 Efforts should be oriented toward the optimization of processes for the production of
51 biosurfactants that result in high productivity on a commercial scale so that they can compete
52 with synthetic surfactants in terms of cost. One of the strategies that has been suggested in the
53 literature (Desai and Banat, 1997; Chayabutra *et al.*, 2000) is the limitation of nutrients, mainly
54 nitrogen sources, as well as multivalent cations and some anions, as a condition for stimulating
55 the synthesis of rhamnolipids by *P. aeruginosa*. In addition to nutritional modifications, changes

56 in physical factors such as temperature and pH can also influence the synthesis of rhamnolipids
57 by *P. aeruginosa*. Guerra-Santos and collaborators (1984) showed that rhamnolipid production
58 peaked at pH values ranging from 6.0 to 6.5 and decreased at values higher than 7.0. Another
59 approach to the induction of the rhamnolipid synthesis is the use of exogenous or endogenous
60 autoinducers (Ochsner and Reiser, 1995; Nakata *et al.*, 1998).

61 The aim of this work was to improve the cultivation conditions for the production of
62 rhamnolipids by a strain of *Pseudomonas aeruginosa* isolated from a Brazilian petroleum-
63 exploring environment. This study involved the investigation of the effect of a variation in pH
64 and the addition of endogenous homoserine lactones, as well as studies on fed batch operation.

65

66 **Material and Methods**

67 **Inoculum**

68 *Pseudomonas aeruginosa* PA1 (Santa Anna *et al.*, 2001) was maintained in a glycerol
69 solution (10% v/v) at -80°C . The thawed strain sample was inoculated onto YPDA plates (yeast
70 extract, 0.3%; peptone, 1.5%; dextrose, 0.1%; Agar, 1.2%) at 30°C for 48 h. The growth of the
71 inoculum was initiated by the addition of a loopful of cells from YPDA plates to a 1000 mL
72 Erlenmeyer flask containing 300 mL of medium with the following composition (per liter):
73 NaNO_3 , 1.0 g; KH_2PO_4 , 3.0 g; K_2HPO_4 , 7.0 g; $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 0.2 g; yeast extract, 5.0 g; peptone,
74 5.0 g and glycerol, 30 g. *P. aeruginosa* was grown at 30°C in a rotary shaker at 170 rpm for 24
75 hours. Cells were harvested by centrifugation (10,000g for 30 minutes) and used as the
76 inoculum.

77

78 **Influence of pH on rhamnolipid production**

79 The culture medium contained (per liter) 0.2 g of $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 1.38 g of NaNO_3 , and 30 g
80 of glycerol. The pH of the culture medium was adjusted from 6.0 to 8.0 with 0.062 M
81 $\text{KH}_2\text{PO}_4/\text{K}_2\text{HPO}_4$ buffer. Six 500 mL Erlenmeyer flasks containing a working volume of 100 mL
82 were prepared, inoculated with 0.12 g of cells (dry weight) and incubated at 30°C in a rotary
83 shaker at 170 rpm for 192 hours. At 24-hour intervals, samples were removed for the
84 measurement of cell growth and rhamnolipid concentration.

85

86 **Simple batch process supplemented with endogenous acylated homoserine lactones (AHL)**

87 The culture medium was prepared with the following composition (per liter): NaNO_3 , 1.38
88 g; KH_2PO_4 , 3.0 g; K_2HPO_4 , 7.0 g; $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 0.2 g and glycerol, 30 g. To each of five 500
89 mL Erlenmeyer flasks was added 100 mL of liquid medium supplemented with 1%, 5%, 10%,
90 15% or 20% (v/v) of cell-free spent culture supernatant from a 120 h-old *P. aeruginosa* PA1
91 culture. The control flask contained no endogenous AHL. The flasks were inoculated with 0.05 g
92 of cells (dry weight) and incubated at 30°C and pH 7.0 in a rotary shaker at 170 rpm for 200
93 hours. Samples were removed at 24-hour intervals for the assessment of cell growth and
94 rhamnolipid production.

95

96 **Fed batch process: nitrogen and carbon feed**

97 The culture medium contained (per liter) 3.0 g of KH_2PO_4 , 7.0 g of K_2HPO_4 , 0.2 g of
98 $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 0.46 g NaNO_3 , and 10.0 g glycerol. Three 500 mL Erlenmeyer flasks (A, B and
99 C) were prepared with 100 mL of liquid medium. The flasks were inoculated with 0.05 g of cells
100 (dry weight) and incubated at 30°C in a rotary shaker at 170 rpm for 240 hours. The frequency of
101 addition was a function of the amount of glycerol consumed in each flask. Flask A received 5
102 mL of a solution containing glycerol (200 g L^{-1}) and sodium nitrate (9 g L^{-1}) at each addition.

103 Flask B received 5 mL of a solution containing only glycerol (200 g L⁻¹), and flask C received 5
104 mL of sterile distilled water. Samples were removed at 24-hour intervals for the quantification of
105 cell growth, nitrate and glycerol consumption and rhamnolipid production.

106

107 **Thin-layer chromatography for rhamnolipid analysis**

108 The rhamnolipids were extracted with ethyl acetate from acidified (pH 2.0) cell-free
109 medium and analyzed by TLC on silica-gel-coated aluminum sheets (Macherey-Nagel®) using
110 CHCl₃:MeOH:AcOH (65:15:2) as the eluent (Schenk *et al.*, 1995). The separated zones were
111 stained with orcinol-sulfuric acid reagent, followed by heating at 100°C for 15 minutes. The
112 plates were scanned, and the relative quantities of the spots corresponding to the
113 monorhamnolipids and dirhamnolipids were determined by densitometry using Band Leader
114 (Ma'ayan Aharoni) and Micronal Origin (Micronal Software, Inc) software.

115

116 **Determination of cell, rhamnolipid, glycerol and nitrate concentrations**

117 Cell growth was assessed by measuring the absorbance at 500 nm, and the cell dry weight
118 (g/L) was determined using a standard calibration curve [ABS = 1.2595 x DW (g/L) – R²=0.989],
119 considered to be valid for absorbance values up to 0.6 OD. Rhamnolipid quantification was
120 achieved indirectly by measurement of the rhamnose concentration using the method of Dubois
121 (1956). A 0.5 mL volume of cell-free supernatant was mixed with 0.5 mL of 5% phenol solution
122 and 2.5 mL of sulfuric acid and incubated for 15 minutes before measuring the absorbance at 490
123 nm. The results were compared with the analytical curve for rhamnose. Glycerol was quantified
124 by the GPO-POD enzymatic-colorimetric method using a kit for triglyceride determination from
125 LaborLab (Brazil). Nitrate was quantified through a colorimetric method using brucine sulfate.

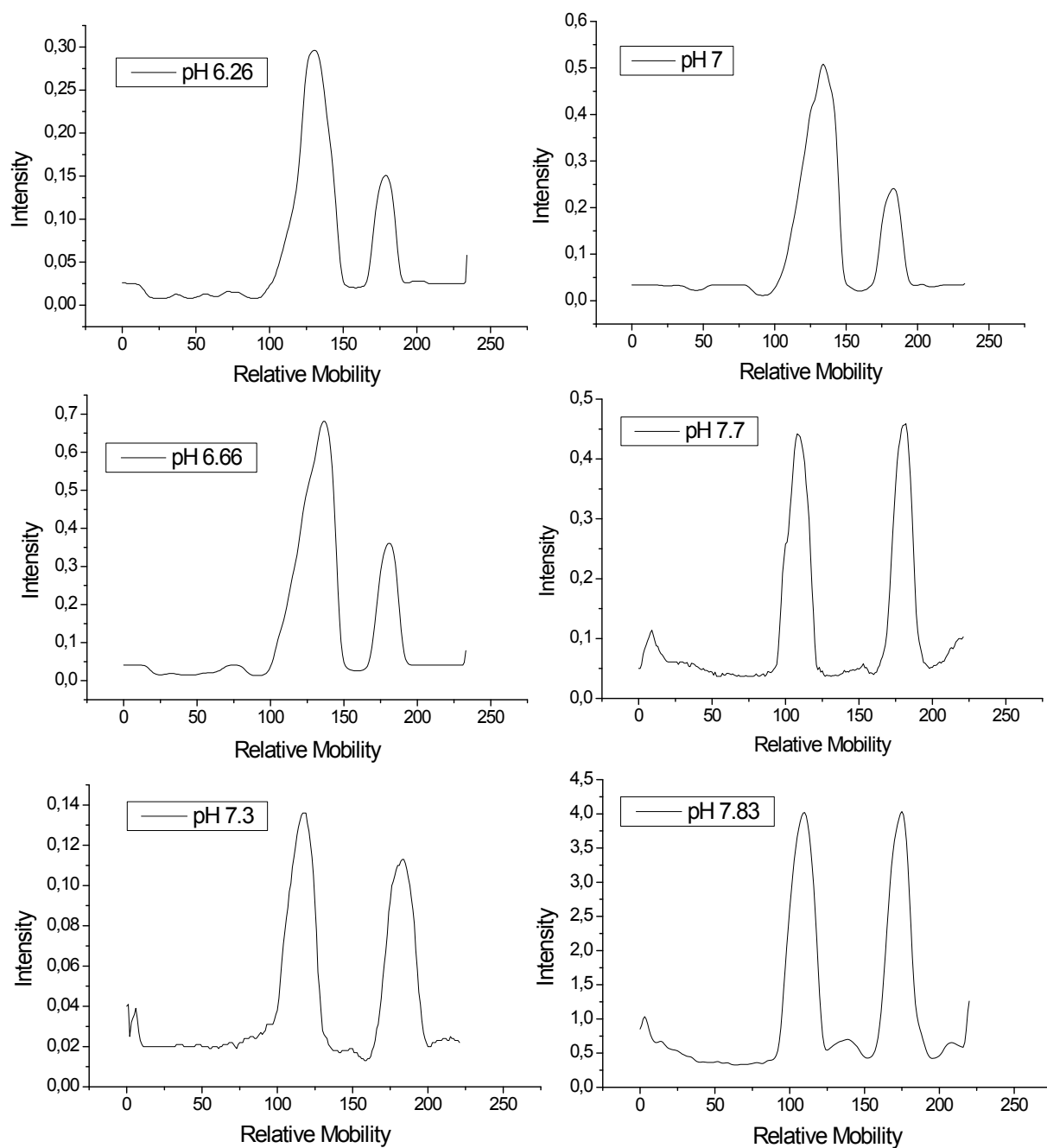
126 Briefly, 2 mL of 0.6 g L⁻¹ brucine sulfate in sulfuric acid solution (80%) was added to 0.5 mL of
127 sample, and the reaction mixture was heated in boiling water for 15 minutes. The reaction
128 mixture was immediately cooled in an ice bath, and the absorbance was measured at 410 nm.
129 The absorbance values were converted into concentration using an analytical curve for sodium
130 nitrate.

131

132 **Results and Discussion**

133 **Influence of pH on rhamnolipid synthesis**

134 The qualitative effect of pH on rhamnolipid synthesis was evaluated by thin layer
135 chromatography. Densitometric analysis of mono- and dirhamnolipids by TLC using (Figure 1)
136 furnished relative migration values close to 0.8 for the monorhamnolipids and 0.5 for the
137 dirhamnolipids. These values are in agreement with the migration rate of monorhamnolipids and
138 dirhamnolipids observed by Schenk and collaborators (1995), who employed TLC analysis under
139 the same conditions.



140

141 **Figure 1:** Densitometric measurements of the thin layer chromatographs of rhamnolipids at
142 various pH levels.

143

144 As shown in the Table 1, the percentage of mono- and dirhamnolipids, determined by
145 densitometry of TLC, varied as a function of the pH of the culture medium. For $\text{pH} \leq 7.0$, the
146 amount of monorhamnolipids produced was lower than the amount of dirhamnolipids. On the
147 other hand, at pH values higher than 7.0, the amount of mono- and dirhamnolipids was
148 approximately the same.

149

150 **Table 1:** Relative percentages of the rhamnolipid types synthesized as a function of the pH of the
151 culture medium.

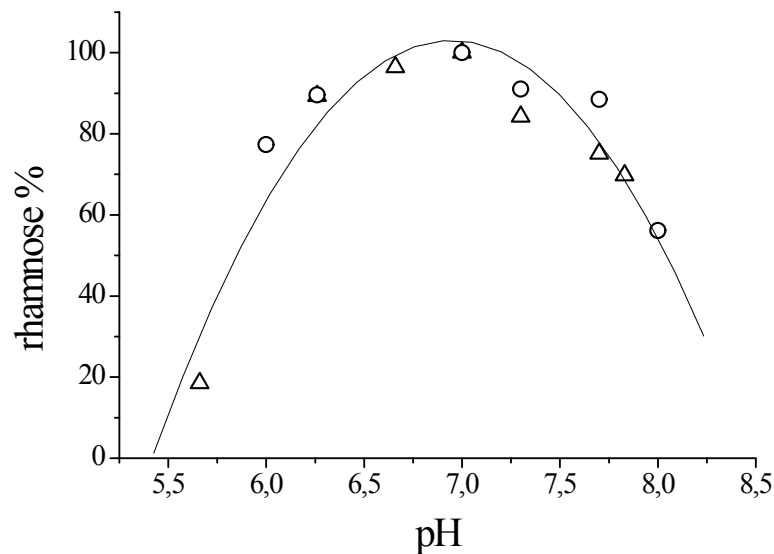
pH	Dirhamnolipids (%)	Monorhamnolipids (%)
6.26	74.35	25.65
6.66	75.35	24.65
7.00	76.41	23.59
7.30	55.25	44.75
7.70	51.66	48.34
7.83	52.75	47.25

152

153 It is possible that pH acts at a transcriptional level on the synthesis of one of the
154 rhamnosyltransferases or on the synthesis of a specific glycolipid transporter channel, or even on
155 the synthesis of L-rhamnose in *P. aeruginosa* (Olvera *et al.*, 1999). *Escherichia coli* represents
156 an example in which various enzymes and periplasmic proteins are expressed in a pH-dependent
157 mechanism (Stancik *et al.* 2002). In any case, the possibility of obtaining a product with distinct
158 percentages of rhamnolipid types synthesized by the appropriate choice of pH allows for
159 different uses and applications for this natural formulation. Indeed, different rhamnolipid types
160 or its mixtures present emulsifying activity, critical micellar concentration, hydrophobicity or
161 bioavailability distinct from each other (Mata-Sandoval *et al.*, 1999).

162 The quantitative effect of the pH of the culture medium on total rhamnolipid synthesis
163 demonstrated that the most productive pH was 7.0 (Figure 2).

164



165

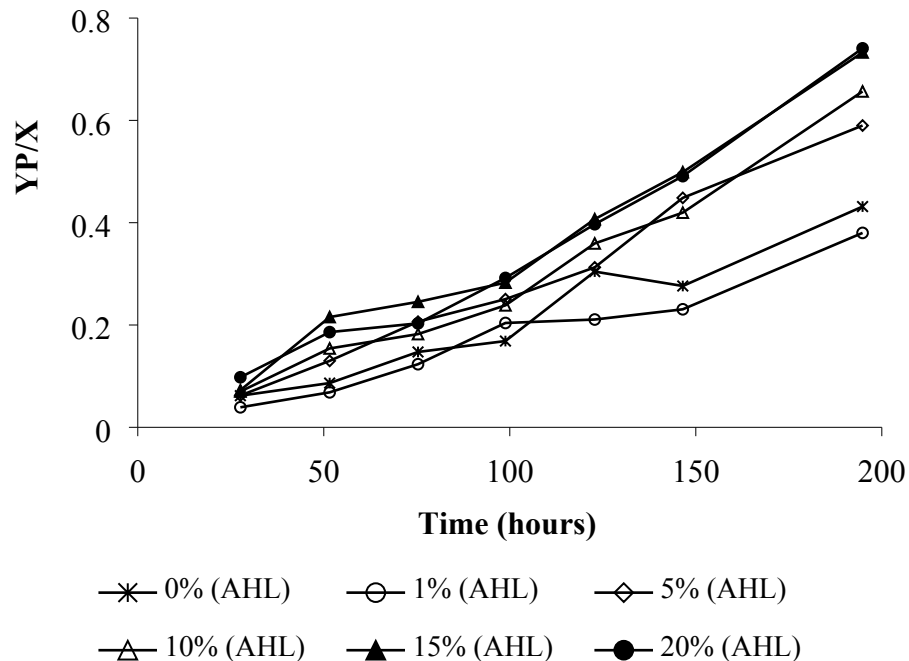
166 **Figure 2:** Variation of rhamnolipid synthesis as a function of the pH of the culture medium. The
167 quantification of rhamnolipids by the Dubois method expresses the rhamnolipid content as the
168 rhamnose concentration. The triangles and circles correspond to independent experiments.

169

170 Use of endogenous autoinducers of the *P. aeruginosa* quorum sensing system

171 On the basis of an approach motivated by knowledge of the regulation of rhamnolipid
172 synthesis by the quorum sensing system in *P. aeruginosa* (Ochsner and Reiser, 1995), an
173 endogenous mixture of undetermined acylated homoserine lactones (AHL) present in the spent
174 cultivation medium where *P. aeruginosa* was grown for a 120-hour period (later stationary
175 phase) was used. The mass balance (carbon and nitrogen sources) present at the start of the
176 culture process was maintained with the same C/N ratio. The spent supernatant added to new
177 cultures operated in simple batch processes was shown to be extremely effective for the

178 induction of rhamnolipid synthesis. The addition of spent medium containing endogenous AHLs
 179 at the start of new *P. aeruginosa* cultures promoted an increase in the specific yield of
 180 rhamnolipids ($Y_{P/X}$) proportional to the volumetric fractions of supplemental spent culture
 181 medium added (Figure 3).



182

183 **Figure 3:** Progress curve of specific yield ($Y_{P/X}$) in culture medium with different supplemental
 184 proportions of spent medium containing endogenous AHL.

185

186 The addition of endogenous AHLs to the *P. aeruginosa* PA1 culture increased the specific
 187 rate of consumption of the carbon source (q_s) and the yield coefficient $Y_{P/S}$ (Table 2). This fact
 188 suggests that the addition of AHLs would also alter the metabolism of nutrients that, in the final
 189 analysis, would be coupled to the synthesis of the precursors for rhamnolipid and AHL
 190 production. The volumetric productivity (Q_p) was also favored by the addition of endogenous
 191 AHLs (Table 2).

192

193 **Table 2:** Process parameters of rhamnolipid production by *P. aeruginosa* PA1 in culture
 194 medium supplemented with different percentages of spent medium containing endogenous AHL.

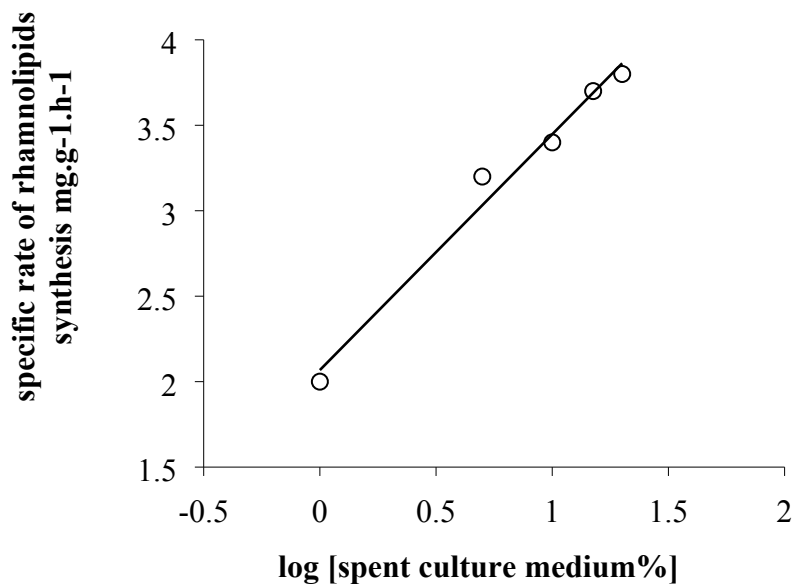
Condition	0%	1%	5%	10%	15%	20%
Δ Rhamnose (g L ⁻¹)	2.15	2.02	3.52	3.63	3.96	4.21
Δ Biomass (g L ⁻¹)	4.98	5.31	5.96	5.53	5.40	5.68
$Y_{P/X}$ (g g ⁻¹)	0.431	0.380	0.590	0.657	0.733	0.741
$Y_{P/S}$ (g g ⁻¹)	0.145	0.149	0.142	0.145	0.164	0.171
Q_P (g L ⁻¹ h ⁻¹)	0.011	0.010	0.018	0.019	0.020	0.022
q_S (g L ⁻¹ h ⁻¹)	0.014	0.012	0.021	0.028	0.029	0.031

195 **Δ Rhamnose:** increase in rhamnose concentration throughout the process; **Δ Biomass:** increase in cellular
 196 concentration throughout the process; **$Y_{P/X}$:** yield factor of product synthesized based on biomass at the end of the
 197 process; **$Y_{P/S}$:** yield factor of product synthesized based on consumed substrate; **Q_P :** volumetric productivity. **q_S :** rate
 198 of glycerol consumption.

199

200 The calculated slopes of each curve shown in Figure 3 (ratio between $\square Y_{P/X}$ and \square Time)
 201 were plotted against the log values of the spent culture medium and are presented in Figure 4.
 202 This plot permits one to clearly observe that the increase in the production of rhamnolipids was
 203 not due to an increase in cellular growth. These results suggest that the amount of AHLs added at
 204 the beginning of cultivation probably contributed to the increase in the synthesis of
 205 rhamnosiltransferases due to the natural unfolding of the quorum sensing system, which is
 206 responsible for the transcriptional regulation of rhamnolipid synthesis (Ochsner and Reiser,
 207 1995).

208



209

210 **Figure 4:** Relation between the specific rate of rhamnolipid synthesis and the addition of
211 exogenous AHLs (expressed in the form of percent volume of culture medium containing
212 endogenous AHL).

213

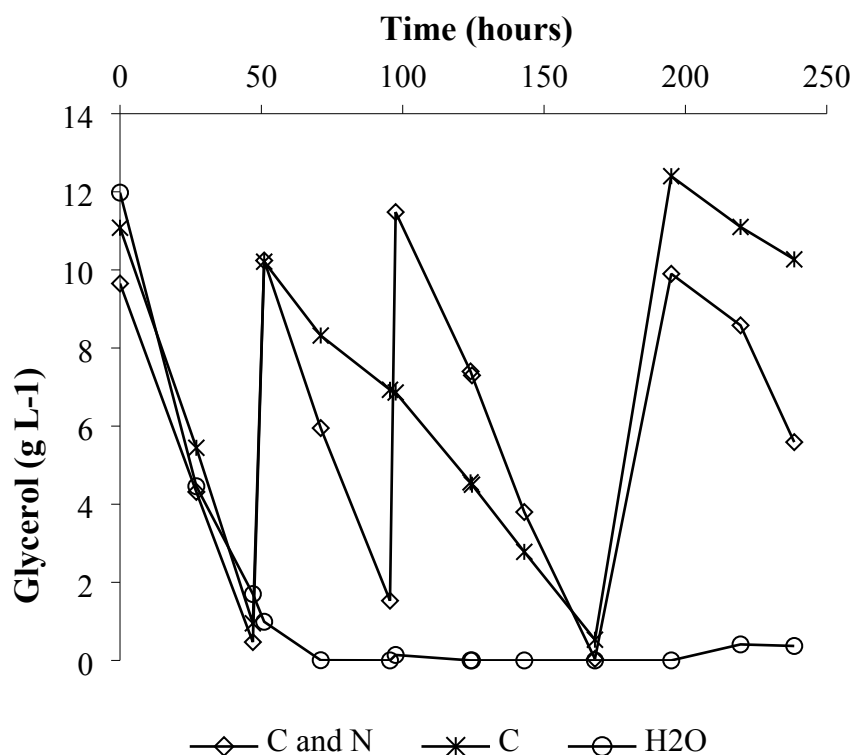
214 **Fed batch strategy**

215 Fed batch operation is another possible strategy to increase the production of
216 rhamnolipids and was chosen to circumvent a possible negative effect, observed by Santa Anna
217 *et al.* (2002), on the assimilation of nutrients when the culture medium contains glycerol
218 concentrations higher than 3%, as well as the inhibition of the cellular growth in cultures of *P.*
219 *aeruginosa* with nitrate concentrations higher than 0.1% (Chayabutra *et al.*, 2000). In addition, a
220 process driven by fed batch can control and maintain the nutrient limits already established as
221 being favorable for rhamnolipid synthesis (Desai and Banat, 1997; Chayabutra *et al.*, 2000).

222 The consumption of nutrients during the fed batch process was determined. In the control
223 experiment, in which only water was fed to the medium, the carbon source (glycerol) and

224 nitrogen source (NaNO_3) present at the beginning were totally consumed after approximately 50
 225 and 24 hours of cultivation, respectively (Figures 5 and 6).

226



227

228 **Figure 5:** Glycerol consumption during the fed batch process using different feeding strategies.

229

230 Although the rate of consumption of the nitrogen source (NaNO_3) was superior to that of
 231 glycerol (Table 3), the addition of a solution containing glycerol and nitrate was accomplished
 232 only as a function of the consumption of the carbon source (Figure 6). The rate of glycerol
 233 consumption was dependent on the presence of a nitrogen source. This fact becomes evident
 234 when the curves of glycerol consumption are compared amongst themselves, considering the
 235 process in which glycerol and nitrate are added together and the process in which only glycerol
 236 was added (Figure 5). It is reasonable to assume that the enzymatic machinery involved in

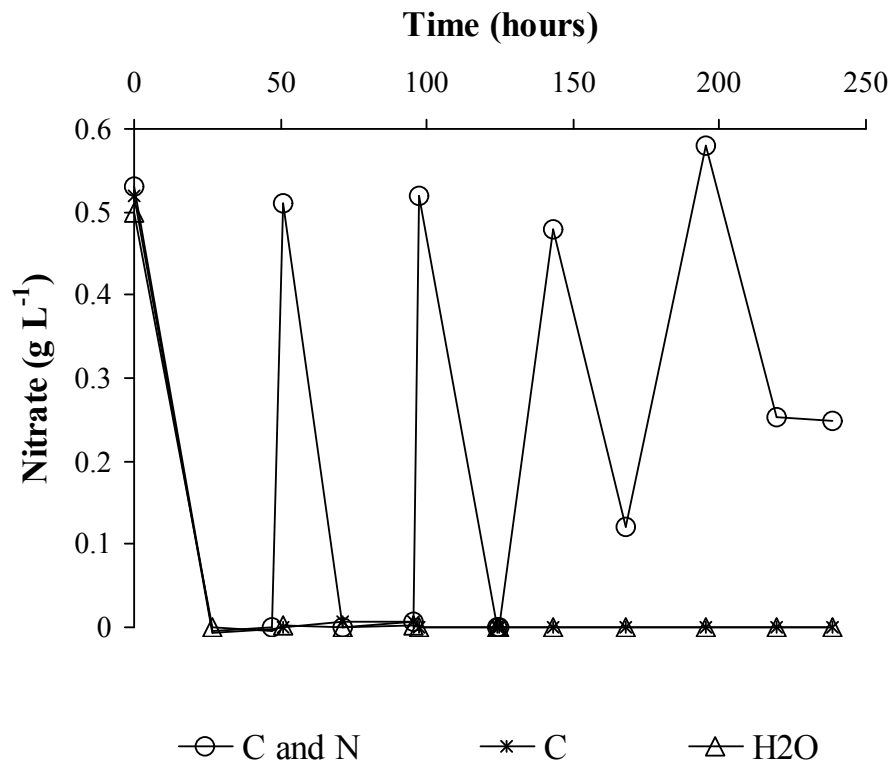
237 glycerol metabolism, as well as in rhamnolipid biosynthesis, depends on the assimilation of
 238 nitrogen and its conversion into catalytic proteins.

239

240 **Table 3:** Process parameters observed in fed batch processes under different feeding conditions.

Conditions	C+N	C	H ₂ O
Δ Rhamnose (g L ⁻¹)	4.90	3.48	1.29
Δ Biomass (g L ⁻¹)	4.78	3.34	2.41
$Y_{P/X}$ (g g ⁻¹)	1.03	1.04	0.514
$Y_{P/S}$ (g g ⁻¹)	0.142	0.168	0.108
Q_P (g L ⁻¹ h ⁻¹)	0.021	0.015	0.013
$q_{S(\text{Gly})}$ (g L ⁻¹ h ⁻¹)	0.168	0.073	0.197
$q_{S(\text{NO}_3)}$ (g L ⁻¹ h ⁻¹)	0.027	0.027	0.027

241 **Δ Rhamnose:** increase in rhamnose concentration throughout the process; **Δ Biomass:** increase in cellular
 242 concentration throughout the process; **$Y_{P/X}$:** yield factor of product synthesized based on biomass at the end of the
 243 process; **$Y_{P/S}$:** yield factor of product synthesized based on consumed substrate; **Q_P :** volumetric productivity; **$q_{S(\text{gly})}$:**
 244 rate of glycerol consumption; **$q_{S(\text{NO}_3)}$:** rate of nitrate consumption. The parameters for the condition “H₂O” were
 245 calculated at 100 h of culture.



246

247 **Figure 6:** Nitrate assimilation during the fed batch process using different feeding strategies.

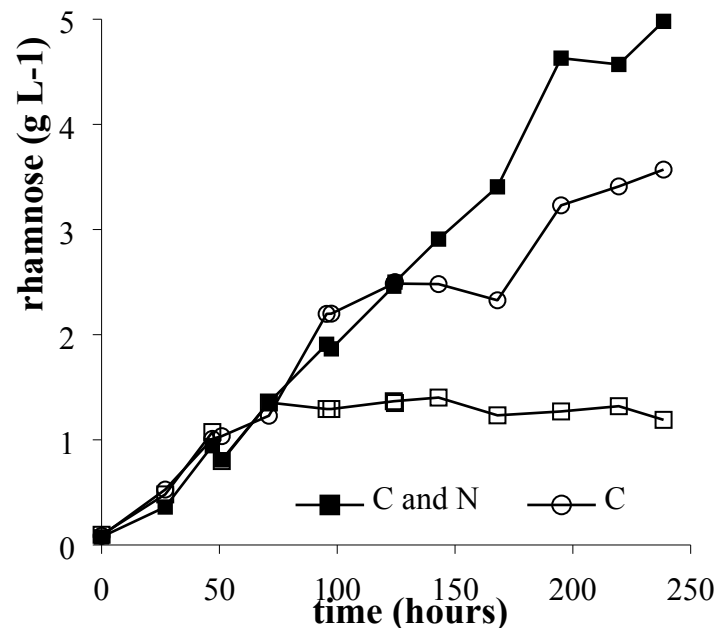
248

249 Ochsner *et al.* (1995) observed that the activity of the rhamnosyltransferase in *P.*
 250 *aeruginosa* during cultivation in a nitrogen-limiting medium containing glycerol as the carbon
 251 source showed maximum activity in the beginning of the stationary phase, which declined to
 252 zero in the late stationary phase. We suggest the hypothesis that exhaustion of the nitrogen
 253 source would not only limit cellular growth, but also the maintenance of the enzymatic
 254 machinery, mainly that involved in the metabolic pathways for rhamnolipid synthesis. When the
 255 fermentation system was fed with carbon and nitrogen sources, a higher cell yield and volumetric
 256 productivity of rhamnolipids was achieved (Table 3).

257

258 On the basis of the feeding conditions, together with the control experiment (without
 feeding), one can conclude that the exhaustion of carbon and nitrogen sources interrupted

259 rhamnolipid synthesis sooner (at 72 hours after initiating the cultivation) (Figure 7). On the other
260 hand, feeding the system with only glycerol led to an improvement when compared with the
261 simple batch (Figure 7) and resulted in higher $Y_{P/X}$ and $Y_{P/S}$ values (Table 3). Furthermore, the
262 feeding condition with both sources of nutrients (carbon and nitrogen) resulted in an even greater
263 increase in both the biomass produced and the volumetric productivity (Table 3).



264

265 **Figure 7:** Time course of rhamnolipid production in the fed batch process using different feeding
266 strategies.

267

268 Conclusion

269 In addition to demonstrating the most appropriate pH for stimulating the production of
270 rhamnolipids, selective synthesis of different types of rhamnolipids caused by certain pH ranges
271 appeared as an unexpected and timely result. The reutilization or recycling of culture medium
272 containing endogenous homoserine lactones (autoinducers of the quorum sensing system)
273 produced by *P. aeruginosa* in new culture medium for the production of rhamnolipids was very

274 effective in the induction of rhamnolipid synthesis. A two-fold increase in volumetric
275 productivity was obtained using this strategy. The fed batch experiment using as strategy of
276 carbon and nitrogen source limitation was successful and could be employed along with an
277 appropriate formulation of the cultivation conditions (micronutrients, pH, supplementary AHLs).

278

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281

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