



Assessing the effect of heavy metals on maize (*Zea mays* L.) growth and soil characteristics: plants-implications for phytoremediation

Muhammad Imran Atta¹, Syeda Sadaf Zehra¹, Habib Ali², Basharat Ali², Syed Naveed Abbas¹, Sara Aimen³, Sadia Sarwar¹, Ijaz Ahmad⁴, Mumtaz Hussain⁵, Ibrahim Al-Ashkar⁶, Dinakaran Elango⁷ and Ayman El Sabagh^{8,9}

¹ Department of Botany, The Islamia University of Bahawalpur, Punjab, Bahawalpur, Pakistan

² Agricultural Engineering, Khwaja Fareed University of Engineering and Information Technology, Rahim Yar Khan, Pakistan

³ Department of Botany, the Women University, Multan, Pakistan

⁴ Faculty of Agricultural Sciences, Bahauddin Zakariya University Multan, Multan, Pakistan

⁵ Department of Veterinary Sciences, The Islamia University of Bahawalpur, Punjab, Bahawalpur, Pakistan

⁶ Department of Plant Production, College of Food and Agriculture Sciences, King Saud University, Riyadh, Saudi Arabia

⁷ Department of Agronomy, Iowa State University, Ames, USA

⁸ Department of Agronomy, Faculty of Agriculture, Kafrelsheikh University, Kafr al-Sheik First, Egypt

⁹ Department of Field Crops, Faculty of Agriculture, Siirt University, Siirt, Turkey

ABSTRACT

Background. Heavy metal pollution has become a global environmental issue. Heavy metals are contaminating the agro-soils, growing crops, and vegetables through different agricultural practices. In this study, besides the phytoremediation potential of maize, the role of chromium (Cr) and lead (Pb) on crop and soil health has been investigated.

Methods. Two maize varieties, Pak-Afgoi and Neelem, were grown under varying concentrations of Cr (50–300 ppm) and Pb (30–300 ppm) and different growth parameters *i.e.*, seed germination, leaf size/number, stem girth, plant height, biomass, chlorophyll content, relative growth rate (RGR), and net assimilation rate (NAR) were studied under Cr and Pb stress. Likewise, the effect of metals was also assessed on different soil characteristics including soil texture, pH, EC, soil organic matter, urease activity and nutrients.

Results. Studied plant attributes were adversely affected by heavy metals toxicity. Affected values of RGR and NAR showed a linear correlation with affected growth and dry matter yield of maize. Heavy metals impacted different soil parameters including soil microbial performance and revealed a declining trend as compared to control soil. Maize varieties showed a significant phytoremediation potential *i.e.*, uptake of Cr and Pb was 33% and 22% in Pak-Afgoi, while Neelem showed 38% and 24% at 300 ppm, respectively. Data regarding metal translocation factor (TF), bioaccumulation factor (ACF), and biomagnification ratio (BMR) significantly revealed the potential of maize varieties in the removal of Cr and Pb metals from affected soils. However, Cr-accumulation was higher in shoots, and Pb accumulated in plant roots showed a differential behavior of metal translocation and affinity with the varieties. These maize

Submitted 16 June 2023

Accepted 20 August 2023

Published 9 October 2023

Corresponding authors

Muhammad Imran Atta,
imranbotany80@yahoo.com

Habib Ali, habib.ali@kfueit.edu.pk

Academic editor

Haider Mahmood

Additional Information and
Declarations can be found on
page 18

DOI 10.7717/peerj.16067

© Copyright
2023 Atta et al.

Distributed under
Creative Commons CC-BY 4.0

OPEN ACCESS

varieties may be recommended for general cultivation in the Cr and Pb-contaminated areas.

Subjects Agricultural Science, Ecology, Plant Science, Environmental Contamination and Remediation

Keywords Plant Science, Bioaccumulation factor, Biomagnification ratio, Net assimilation rate, Phytoremediation, Environmental contamination and remediation, Agricultural science, Molecular biology

INTRODUCTION

Abiotic stresses including metal toxicity, salinity, temperature extremes, soil microplastic, and drought are enormous threats that are affecting agriculture and the natural environment (Wang, Vinocur & Altman, 2003; Hasanuzzaman et al., 2020; Li et al., 2023). Heavy metals are one of the abiotic factors and are demarcated as metals with a density above 5 g/cm³ e.g., chromium (Cr), lead (Pb), nickel (Ni), cobalt (Co), arsenic (As) and silver (Ag), etc. These heavy metals diverge in physical and chemical properties; and are taken as substantial environmental pollutants owing to their toxic interaction with soil properties, plants, animals, and humans (Hasan et al., 2009; Das & Jayalekshmy, 2015; Zhang et al., 2019; Karkush & Ali, 2020). An incessant boost of heavy metals in the agricultural soil system is over diverse agricultural practices including the use of industrial and sewage waste-waters as a crop irrigation source. Irrigation of vegetables and fodder crops with such a kind of wastewater through their discharge into freshwater bodies is a common practice (Khan, Khan & Aslam, 2003); and hence is the foremost source of heavy metal pollution for intact growing crops in the peri-urban areas (Mussarat, Bhatti & Khan, 2007).

Chromium (Cr) is a toxic pollutant and is ranked as the 17th most toxic element among the hazardous substances (Wakeel, Xu & Gan, 2020). Its density is 7.15 g/cm³ and ranged from 10 to 50 mg/kg of soil, naturally (Kouser & Khan, 2021). Chromium is used in electroplating, textile dyeing, paint, metallurgy, pigment, and tanning industry. Similarly, sewage and fertilizers are also included as the main source of Cr (Amin et al., 2013). Due to different oxidation states, Cr acts as a toxic element for organisms. However, Cr-toxicity to plants depends on its uptake mechanisms, concentration, and focal plant species (Gardea-Torresdey et al., 2005; Peternella, da Silva & da Costa, 2021).

Lead (Pb) is ranked as the second most toxic metal on earth's crust and is toxic to humans and other living things including plants. Its density is 11.34 g/cm³ (World Health Organization, 2010). Several industrial processes include Pb-use in their products like oil and paint, mines, agrochemicals, etc. Moreover, Pb as salts or oxides is also being added to the environment through atmospheric dust, and automobile exhaust (IARC, 2012; Kumar et al., 2019). In nature, Pb remains below 50 mg kg⁻¹, but in some plants, Pb usually inhibits the growth mechanism when it is at a concentration of 30 mg/kg or more (Usman et al., 2020), while some of the plant species can tolerate Pb stress up to 1,000 mg kg⁻¹ (Reeves et al., 2018). Both Cr and Pb have a strong effect on different growth attributes of

exposed plants (wheat, maize, barley, sunflower, mustard, and soybean); and inhibit seed germination, plant height, root-shoot length, fresh-dry weight of seedlings, tolerance index, leaf number and photosynthesis (Orhue & Ekhomun, 2010; Naseem et al., 2015; Akhtar & Iram, 2017; Kanwal et al., 2020).

Changes in soil properties depend upon the mobility and chemical activity of heavy metals in the soil predominantly when these metals exceed the accepted limits (Karkush, Zaboon & Hussien, 2014; Uddin, 2016; Karkush & Ali, 2020). Metal ions have acidification effects on the intact soil and lower the pH of the soil (Motuzova, Makarichev & Petrov, 2011). Soil pH is an important parameter that significantly affects the accessibility of soil nutrients available to the growing crops, affects their yield, and hence, acts as the key factor in sustainable agriculture (Ludwig et al., 2001; Najafi & Jalali, 2016). Soil respiration (CO₂ evolution) is an indicator of the use of energy by soil microbes concerning their efficiency in degrading the soil organic material (Wardle & Ghani, 1995).

In agricultural management, microbes are taken as soil indicators for the affecting external abiotic stresses including heavy metals (Hassan et al., 2013c), and are quite sensitive to such stresses. Microbes release important extracellular enzymes in the soil system, which are the key regulators of soil biochemical processes (Wang & Yanli, 2013). Soil urease is one of the most concerning extracellular enzymes released by microbes to hydrolyze soil urea into CO₂ and ammonia (Gulser & Erdogan, 2008). Similarly, soil enzymes act as biological catalysts and facilitate different soil reactions and metabolic processes of the biogeochemical cycles of soil nutrients to maintain soil fertility for growing crops (Moreno, Garcia & Hernandez, 2003). As the heavy metals put adverse effects on soil properties, some efficient and cost-effective techniques are needed to restore the metal-affected agro-soils. Phytoremediation is a biological remediation technique that has received a lot of attention during the last few years. Nevertheless, plant efficiency for phytoremediation depends upon the type, availability, and concentration of heavy metals. Phytoremediation easily removes metal contaminants from the affected soil than other remediation options (Marques, Rangel & Castro, 2009). It increases soil fertility through the release of different organic matter (from plant body) and hence, maintains the physical and biological properties of the soil (Aken, Correa & Schnoor, 2009; Wuana & Okieimen, 2011; Jacob et al., 2018). Plants used in the phytoremediation of heavy metals may be hyper-accumulator or phyto-stabilizer. Family Brassicaceae (*Alyssum bertolonii*; *Thlaspi caerulescens*) and Asteraceae (*Calendula officinalis*; *Tagetes erecta*) have a greater hyper-accumulating ability (Glick, 2012). Similarly, the phytoextraction potential of soybean (*Glycine max* L.) and rice (*Oryza sativa* L.) for phytoextraction of cadmium-polluted lands has also been reported by Murakami, Ae & Ishikawa (2007). Moreover, *Lolium perenne*, *Panicum aquaticum*, *Typha* species, *Vetiveria zizanioides*, and *Paspalum fasciculatum* have also been documented as good phytoremediation tools for Cd, Cu, As, Zn, Cr, and Pb (Glick, 2012; Alvarenga et al., 2009; Andra et al., 2009; Dipu, Kumar & Thanga, 2012; Pires-Lira et al., 2020).

Data were collected as previously described by Atta et al. (2023) that due to excessive irrigation with wastewater, Cr and Pb have become the most frequent and health risk metals to the consumers of the study area (Dera Ghazi Khan, Punjab-Pakistan). Similar findings about heavy metal pollution in Dera Ghazi Khan have been reported by Rafique et

al. (2016). To our knowledge, the toxicity of Cr and Pb on soil physicochemical properties and soil enzyme activity under maize cultivation has not been documented adequately in the study area. Therefore, a pot experiment was set to understand the effect of heavy metals not only on *Zea mays* seedlings but also on different soil characteristics. Moreover, this study also uncovered the phytoremediation potential of maize to combat the heavy metal issue in the future under the particular environmental conditions of the area. The maize crop is exceptionally grown as a fodder crop in this study area; therefore, maize has been selected for the current phytoremediation study.

MATERIALS & METHODS

Experimental design, germination and growth attributes

To evaluate the metal toxicity and phytoremediation potential of maize varieties, separate Petri plate and pot experiments were conducted in a block design during July–September 2018–20 (Temp. 35 °C–32 °C, Humidity 56%–59%). Open-pollinated varieties (OPV) of maize (Pak-Afgoi & Neelem Desi) were used in the trial. Potassium dichromate and lead nitrate ($K_2Cr_2O_7$) & $Pb(NO_3)_2$ were used and a stock solution was prepared *viz.* Cr (50, 100, 150, 250, 300 ppm) and Pb (30, 60, 100, 150, 300 ppm).

Each treatment comprised of eight replicates followed by three maize plants per treatment.

A seed germination test was performed in the laboratory at room temperature (30 °C). To prevent fungal infection during the experiment, the selected seed material was thoroughly washed with 2% sodium hypochloride for 5 min and then rinsed with distilled water. Seeds were imbibed in distilled water for 30 min and then were air-dried. For either variety, each Petri plate (10 cm in diameter) was employed with two filter papers and 10 seeds following four replicates per treatment. Each Petri plate was moistened with 10 mL of the metal solution while the control treatment continued with distilled water. Overall, the plates were observed daily for moisture/treatment requirements. Germinated seeds with one mm radicle were counted daily till the final germination day (day 10). Percent seed germination was determined following the study of *Akinci & Akinci (2010)* using the formula:

(Germination (%) = Total seeds germinated / total seeds arranged \times 100).

Pre-washed, cleaned, dried, and labeled plastic pots of varying identifiable colors (dimension (cm): 30.5 diameter \times 46 deep) were smoothly filled with 12 kg of the agro-soil (clay 63.1%, sand 29.7%, saturation 54%, EC (mS/cm) 2.8, pH 7.6, SOM content 2.7%, available-P 8.2 ppm, available-K 182 ppm, and 2.8% N). For either variety, healthy seeds of uniform size were sown 1–2 inches deep in the topsoil of the pots. After ten days of establishment, the seedlings were thinned by removing weak seedlings, and metal treatment was simulated for up to four weeks. Different growth attributes *i.e.*, plant length, leaf area (*Aliu, Fetahu & Rozmam, 2010*), SPAD value, plant fresh and dry weight, relative growth rate (RGR), and net assimilation rate (NAR) were assessed for the varieties at Harvest-1 and Harvest-2 *i.e.*, 25th and 40th day of growth (*Table 1*). RGR and NAR were assessed by using the method of *Causton & Venus (1981)* by the given formula:

$$RGR = \text{Log}w_2 - \text{Log}w_1 / t_2 - t_1$$

Table 1 Comparison of growth at harvest-2 in Pak-Afgoi and Neelem under Cr & Pb treatment.

SOV	Cr (ppm)						Pb (ppm)					F-value	LSD (5%)
	0	50	100	150	250	300	30	60	100	150	300		
Control													
Variety: Pak-Afgoi													
Germination (%)	97.5	97.5	86.5	75	62.6	31	90	82.5	62.5	52.5	35.8	41.8	7.19
Plant height (cm)	68	68.4	66.4	62.1	58.9	53.3	69	66.2	61.7	57.3	49.7	4.86	2.57
Leaf area (cm ³)	74	74	72	69.5	67.5	65.5	72	70	67.8	67.5	63.2	3.56	0.79
Leaf area (cm ³): 25d	31.4	32	30.4	27.8	22.2	22	31	30	27.8	26.2	24.8	2.65	1.45
Green leaf count	10	10	10	9	9	9	10	9.5	9.5	9	9	0.41 ^{NS}	0.17
Fresh weight (g)	46.2	45.8	45.5	43.7	41.4	40.4	45.9	44.8	42.6	39.9	38.4	2.7	1.42
Dry weight (g)	17	17	14.5	13	12.2	10.5	17	14	12.8	12.1	10	3.46	0.47
Dry weight (g): 25d	7.4	7.4	7.4	6.5	6.2	5.5	7.4	7.4	6.2	6.2	5.3	2.18	0.23
RGR (g day ⁻¹)	0.64	0.6	0.5	0.4	0.38	0.32	0.64	0.47	0.44	0.37	0.31	3.05	0.03
NAR	0.29	0.29	0.28	0.24	0.24	0.23	0.31	0.27	0.22	0.18	0.15	2.84	0.03
Leaf chlorophyll (SPAD value)	47	46.3	46	43.1	39.4	34	46	44.3	43	37.2	34.5	3.29	2.51
Variety: Neelem Desi													
Germination (%)	95	85	78	62.5	50	27.5	85	70	62.5	47.5	27.5	38.9	6.31
Plant height (cm)	54.1	53.9	51.2	47.7	43	36.9	52.6	48.6	43	39	36.2	9.54	2.78
Leaf area (cm ³)	69.9	65.6	60.8	56.4	51	49	64	60.2	55.2	50.4	48.2	8.03	2.83
Leaf area (cm ³): 25d	29.3	29	26.4	23.5	21	20	29.3	26.4	23.2	21.6	20	2.88	2.03
Green leaf count	8	8	7	7	7	7	8	7.5	7	7	7	1.98	0.17
Fresh weight (g)	36.6	36.4	35.7	33.4	29.5	25.9	36.6	35.5	32.6	28.8	24.8	2.11	1.94
Dry weight (g)	14	13.2	11.5	10	9	8	13	11	9.8	9	7.7	13.8	0.58
Dry weight (g): 25d	6.8	6.9	6.4	5.5	5	4.3	6.9	6.4	5.5	4.8	4.3	9.44	0.35
RGR (g day ⁻¹)	0.48	0.43	0.35	0.3	0.26	0.24	0.42	0.32	0.29	0.26	0.21	3.52	0.02
NAR	0.18	0.16	0.15	0.14	0.14	0.13	0.15	0.14	0.14	0.13	0.11	3.73	0.01
Leaf chlorophyll (SPAD value)	46.1	46	45	41.4	37.3	34.7	44.6	42.5	39.7	35.7	33.2	4.19	1.97

Notes.

NS, statistically not significant.

where, w_2 = plant dw at harvest time of 40 d (t_2), w_1 = plant dw at harvest time 25d (t_1)

$$\text{NAR} = 2(w_2 - w_1) / (LA_1 + LA_2)(t_2 - t_1)$$

where, w_2 = leaf dw at harvest time 40 d (t_2), w_1 = leaf dw at harvest time 25 d (t_1), LA_1 = leaf area measured at harvest time 25 d (t_1), LA_2 = leaf area measured at harvest time 40 d (t_2).

Determination of soil parameters

Soil texture (including clay 63%, sand 30%) was determined with the Bouyoucos hydrometer method by preparing a soil paste that was saturated with distilled water (Sheldrick & Wang, 1993). Soil pH (H₂O) and EC (mS/cm) were determined using a pH and EC meter, respectively. For this purpose, soil-water suspension 1:2.5 (w/v) was prepared, and the cathode of the meters was dipped into it (Hassan et al., 2013c).

Determination of SOM

Soil organic matter content was determined by the method of [Walkley & Black \(1934\)](#). For this purpose, reduction of Cr ion by soil organic matter and an unreduced $\text{Cr}_2\text{O}_7^{2-}$ was measured. A total of 0.5 g ground and sieved soil mixed with 10 mL $\text{K}_2\text{Cr}_2\text{O}_7$ (1M) followed by the addition of 20 mL conc. H_2SO_4 . The sample was well shaken for 30 min and the final volume was raised to 200 mL by distilled water. Afterward, soil material was titrated against acidified 0.5 M ammonium ferrous sulfate. Reading of the sample was manipulated from blank upon the appearance of a green endpoint.

Determination of soil urease activity

Soil urease activity (UA) was determined by the method of [Kandeler & Gerber \(1988\)](#) as described by [Hassan et al. \(2013c\)](#). For this purpose, metal-treated 5 g soil was mixed with 10 mL of urea solution; and then 10 mL of buffer solution (citric acid, KOH, and NaOH) having pH 6.7 was also added to it. This solution was incubated at 37 °C for 24 h. After filtration, the solution was mixed with reagents (phenol + NaOH); to this solution, sodium hypochlorite solution was also added. The absorbance of the appeared blue color was noted at 578 nm through a spectrophotometer.

Determination of soil respiration

For the determination of soil (microbial) respiration, a laboratory incubation experiment was performed to measure soil respiration under two different heavy metals following the method of [Anderson \(1982\)](#) as described by [Devi & Yadava \(2009\)](#). Soil samples were moistened with the five respective doses of either metal and were placed in closed jars provided with test tubes containing NaOH and distilled water test tubes. Evolved carbon dioxide over time was trapped by NaOH titrated with the acid of known normality.

$$\text{(formula: mg of CO}_2 = V \times N \times 22)$$

where V = volume of acid used against 10 ml NaOH N = normality of acid used; and the value 22 is a factor for CO_2 evolved during reaction.

Estimation of soil nutrients (N, P, K)

Soil K^+ was assessed by flame photometer taking a soil sample (2.5 g) by shaking with 33 mL of 1M KCl following [Anderson & Ingram \(1993\)](#). The excess K^+ in the soil sample was washed three times with 95% ethanol and the adsorbed K^+ was then extracted by addition of 33 mL of 1M NH_4OAc . The volume of this extract was raised to 100 mL and further added with 1M NH_4OAc to estimate K^+ in the extract. Similarly, soil phosphorus was assessed following the method described by [Olsen & Sommers \(1982\)](#). For available phosphorus, 5 g of soil was obtained in a 250 mL flask, and 0.5M NaHCO_3 (100 mL) was added to it. This solution was shaken for 30 min and then the filtrate was collected. Ten mL of the filtrate was shifted into a flask of 50 mL along with 1 ml of 5N H_2SO_4 (sulfuric acid) and the volume was increased up to 40 mL by adding distilled water. To this solution, 8 mL of ascorbic acid as a reagent was added to develop color; and transmittance was recorded at 880 nm using a spectrophotometer. Soil nitrogen was determined by Kjeldahl's method

(*Ahmad et al., 2011*) using the formula:

$$N(\%) = \frac{\text{acid used for sample} - \text{acid used for blank} \times \text{acid normality}}{\text{volume of sample}} \times 14.01 \times 10 \times 100.$$

Metal detection in soil and plants

For determination of soil metal content, maize plants were separated from contaminated pot soil was taken out from the respective pots. These samples were executed following the method of *Welz & Sperling (1999)* using an atomic absorption spectrometer (Perkin-Elmer). Hot acid digestion was used for a 1 g soil sample using 15 mL of the acid mixture in a 5:1:1 ratio (70% HNO₃, 70% H₂SO₄, and 65% HClO₄). After cooling, the transparent acidic solution was filtered (Whatman no. 42) and diluted with distilled water. Metal analysis was carried out at analytical spectral lines *i.e.*, Cr: 357.9 nm, and Pb: 283.3 nm. A similar digestion procedure was executed for plant metal detection. For metal accumulation and translocation study, different plant parts (root, stem, leaves, *etc.*) were used. In the treated plants, the bio-magnification ratio (BMR) and metal accumulation factor (ACF) was assessed by the method of *Baker et al. (1995)* whilst the metal translocation factor (TF) was calculated according to *Yanqun et al. (2005)* using the following equations:

$$\text{BMR} = \text{PU/MA}$$

$$\text{ACF} = \text{PU/MT}$$

$$\text{TF} = \text{Element (shoot)/Element (root)}$$

where, PU = metal concentration in whole plant ($\mu\text{g g}^{-1}$), MA = available metal concentration in soil ($\mu\text{g g}^{-1}$), MT = total metal concentration in soil ($\mu\text{g g}^{-1}$)

Quality control analysis and assurance

Chemical analysis of samples was performed by AAS and spectrophotometer. High grade standard chemicals and glass ware were used (Merck-Germany). By using a calibration curve, calibration of instruments was executed with a series of standard solutions of varying concentrations. The chemical stock solution was prepared with double-deionized water. Glass ware was used after cleaning and rinsing with diluted HNO₃ to avoid some probable contamination. For quality results, each sample was analyzed in a repeated way by following the standard reference procedure (*Atta et al., 2023*).

Data analysis

For comparison of the significance level of means under metal treatment, analysis of data was performed by calculating the F-value from ANOVA test using a statistical package IBM-SPSS (V. 20). While error graphs (LSD 5%) were prepared in MS-Excel.

RESULTS

Effect of Cr and Pb on plant growth-related parameters

Seed germination

The Cr and Pb treatments suppressed the maize seed germination in a concentration-dependent manner (Table 1). The decrease in seed germination of both varieties was much more obvious at 150 ppm Cr and 100 ppm Pb treatment. At the highest concentration (300 ppm), Cr and Pb inhibited the seed germination of Pak-Afgoi by 67% and 64% and Neelem by 68% and 73%, respectively.

Green leaves and leaf area

Table 1 shows a minor suppressive effect of Cr and Pb at the early growth stage of maize varieties. Green leaf count was not significantly affected in Pak-Afgoi as compared to the variety Neelem. Cr rapidly decreased this agronomic trait at 150–300 ppm (Pak-Afgoi 5–10%; Neelem 6–29%) and 100–300 ppm Pb application (Pak-Afgoi 5–10%; Neelem 18–29%). A decline in leaf count was more at 300 ppm of Cr and Pb. Likewise, leaf area was also decreased along with the increasing metal doses in a more significant way, and decreased much at elevated levels of Cr and Pb (300 ppm), whereas Neelem declined more than Pak-Afgoi (30 and 31%). Likewise, leaf area was also measured at 25th d (harvest-1) which decreased by 21–29% in Pak-Afgoi, while a decrease in Neelem was up to 32% under Cr and Pb stress.

Plant height

Pb has more adverse effects on plant height than Cr. Comparatively, the Neelem variety showed a pronounced decreasing trend for this agronomic trait. During the early growth stage, the maximum plant length for Pak-Afgoi and Neelem was recorded up to 68 cm and 54.1 cm, respectively. Plant height decreased much at higher metal concentrations (300 ppm). Plant height decreased in Pak-Afgoi under Cr and Pb by 22% and 23%, respectively. Plant height also decreased in Neelem by 32% and 45%, respectively (Table 1). Both the varieties showed a tolerant behavior and were least affected at 50 and 30 ppm of Cr and Pb, whilst rapidly declining at 150 ppm Cr and 100 ppm Pb.

Shoot girth

Both Cr and Pb affected the shoot girth of maize varieties in a declining and concentration-dependent pattern. During the early growth stage, the maximum shoot girth of Pak-Afgoi and Neelem was 6.2 cm and 5.5 cm in the control treatment, whilst the mean decrease in shoot girth of Pak-Afgoi *viz.* varying concentration of Cr and Pb metals was 11–11.3% and 13%, respectively. However, at 300 ppm of Cr and Pb; the mean decrease was more than 19% (Pak-Afgoi) and 27% (Neelem). Maize variety Neelem was less affected than Pak-Afgoi for this circumference trait (Table 1).

Plant biomass

Plant fresh weight and dry weight were assessed under two different metals that revealed metal-induced toxicity on plants. Statistical analysis has predicted a less significant effect of metals on the fresh weight of Pak-Afgoi than of Neelem. Comparative to the control value

(46 g) and to the low concentrations of Cr and Pb, fresh weight decreased more up to 17% at 300 ppm. Fresh weight of Neelem in the control treatment was 26.6 g later decreased up to 27% by Cr and 32% by Pb, likewise Pak-Afgoi (Table 1).

Plant dry weight also decreased both at harvest-1 and harvest-2 (Table 1). Pak-Afgoi attained a maximum of 16.2 g dry weight that decreased up to 38% (Cr, Pb). Similarly, Neelem attained 14g dry weight as the control value and later underwent a significant decrease due to metal toxicity at 300 ppm of Cr (43%) and Pb (45%). Overall, data shows a rapid decrease in plant biomass parameters observed from 150 ppm (Cr) and 100 ppm (Pb) than at lower metal concentrations.

Relative growth rate (RGR) and net assimilation rate (NAR)

Data about RGR (g d^{-1} increase in dry matter) revealed that both maize varieties continued growing *viz.* control and metal treatments and gained dry matter between the two harvests (H2-H1). However, a mean decrease in plant growth rate was noted by 28% & 31% (Pak-Afgoi), 31% & 37% (Neelem) under Cr and Pb, respectively. At maximum dose (300 ppm) of Cr and Pb, RGR decreased by 48% & 49% in Pak-Afgoi, while 53% & 56% in Neelem, respectively.

The decreasing trend of RGR and dry matter of plants strongly showed an affected accumulation of metabolites/photosynthate between the two harvests due to metal stress. A similar decreasing trend was observed in the case of NAR. Metal treatment has revealed increased metal toxicity from H1-H2 (as the plant spent more under stressful conditions). A mean decrease in NAR under the Cr effect in Pak-Afgoi and Neelem was 10% & 17%, whilst Pb affected this parameter by 21% & 22.2%, respectively. The decline in RGR and NAR was elevated at elevated metal concentrations (Table 1).

Effect of Cr and Pb on soil physicochemical properties

Soil pH

Soil pH of the control soil was 7.8, lowered up to 6.7 and 6.5 under Cr and Pb application at H2 (day 40), respectively. Results showed a mild effect of metals on soil pH up to 100 ppm Cr and 60 ppm Pb during this incubation period. The effect of metals increased and lowered soil pH more at elevated metal doses (Table 2).

SOM content and urease activity

The control value of SOM at H2 was 2.8%, whilst a rapid decrease in SOM content was initiated at 150 ppm Cr and 100 ppm Pb. The mean cumulative decrease in SOM content was 17% & 20% under Cr (50–300 ppm) and Pb (30–300 ppm) toxicity. At 300 ppm Cr and Pb application, SOM content decreased by 43% and 46%, respectively (Table 2). Soil urease activity (UA) was found affected by different concentrations of the metals. At harvest time (day 40), the mean decrease in UA was up to 22% & 26% due to Cr and Pb, respectively. Table 2 shows that enzyme activity was much more affected at the elevated metal concentrations than at lower. Although lower metal doses had the least effect on UA, a clear decline in UA was initiated at 100 ppm level of the metals which turned to its peak at 300 ppm *i.e.*, decreased under Cr (41%) and Pb (47%).

Table 2 Assessment of some soil parameters at harvest-2 (40th day) under Cr & Pb stress.

SOV	Cr (ppm)						Pb (ppm)					F-value	LSD (5%)
	0	50	100	150	250	300	30	60	100	150	300		
pH	7.6	7.4	7.4	7.2	6.6	6.4	7.2	7	6.7	6.4	6.3	2.5	0.1
SOM (%)	2.8	2.6	2.46	2.31	2	1.8	2.7	2.51	2.36	2	1.8	19	0.14
Urease Activity (mg NH ₄ -N kg ⁻¹ 24 h ⁻¹)	14.6	14	13.2	11.1	9.7	8.5	14.2	12.5	10.7	9.2	7.6	15.3	1.01
CO ₂ evolution (mg):	176	171	165	153.4	145	124	177	171.3	160.7	149.1	135	12.7	6.54
Soil nutrients													
N (%)	2.82	2.5	2.3	2.1	1.9	1.62	2.4	2.2	2	1.6	1.45	22.1	0.47
P (ppm)	8.2	8.2	8	7.7	7.3	6.6	8.1	8	7.7	7.3	6.3	2.9	0.18
K (ppm)	182	178	172.4	155	142.1	124.1	179.2	177	160.2	143	128.6	9.6	9.03

Soil respiration: (evolution of CO₂)

Table 2 shows the impact of soil metals on the amount of soil CO₂ under cultivation of maize varieties. This parameter of the soil decreased from 176 mg to 124 mg. The mean effect of metal treatments (Cr 50–300 ppm; Pb 30–300 ppm) showed metal toxicity following a significant decrease in soil respiration (SRP), compared to control. SRP decreased more by Cr (14%) than Pb (9.5%) showing much Cr-toxicity on SRP. However, differentiating the treatment effect, 300 ppm Cr and Pb level imposed drastic effects on this parameter by 29% & 23%, respectively.

Soil nutrients (NPK)

Both Cr and Pb treatments have toxic effects on available soil nutrients. Soil nitrogen (N), phosphorus (P), and potassium (K) contents were decreased gradually at lower metal concentrations than at maximum levels. Comparing the toxicity of Cr and Pb on NPK contents at harvest (day 40), the mean decrease in N, and P was more under Pb stress (31%, 7%) than Cr stress (25%, 7%). Likewise, K⁺ was less affected under Pb stress (13%) than Cr (15%), respectively. The decreasing order of NPK at the maximum metal concentration of Cr and Pb was 42%, 17%, 32%, 47%, 22%, and 29%, respectively (Table 2).

Metal uptake from soil (bioaccumulation in plant tissues)

At harvest-2, plant parameters showed a substantial effect of metal toxicity along with the increasing metal concentrations. Cr accumulation was more in the stem than roots and leaves, whilst Pb accumulated more in the roots than stems and leaves of the test varieties (Fig. 1). In Pak-Afgoi and Neelem, maximum Cr was accumulated in the stem (61.2 µg/g & 68.5 µg/g) at its highest concentration in the soil medium. Similarly, Pb accumulation in the stem part of Pak-Afgoi and Neelem was observed at 17.2 µg/g and 19.2 µg/g, respectively. However, Pb contents in the roots were 47.3 µg/g in Pak-Afgoi and 51.2 µg/g in Neelem. Although, metal accumulation in different plant parts was increasing way, however, a hasty metal up taken by plants occurred at 150 ppm (Cr) and 100 ppm of Pb (Fig. 2).

The phytoremediation potential of maize varieties was assessed by calculating the metal accumulation factor (ACF), translocation factor (TF), and bio-magnification ratio (BMR). Results clearly showed significantly increased values of Cr and Pb metals for ACF, TF, and BMR by variety Neelem than Pak-Afgoi. However, both the varieties significantly removed Cr and Pb content from the soil and accumulated in different parts successfully (Fig. 3).

DISCUSSION

Results from the Petri plate experiment have shown that both the tested varieties undergo abiotic metal stress and seed germination decreased with the increasing concentration of Cr and Pb. Studies on seed germination characteristics showed its inhibition under metal (Pb) toxicity even at low or micro-molar levels (Kopittke, Asher & Menzies, 2008). However, there are few reports about the progression of seed germination and inhibition of radical/hypocotyl length in *Elsholtzia argyi* (Islam et al., 2008), but the same was not observed in the present course of the investigation. All the concentrations of Cr and Pb were found

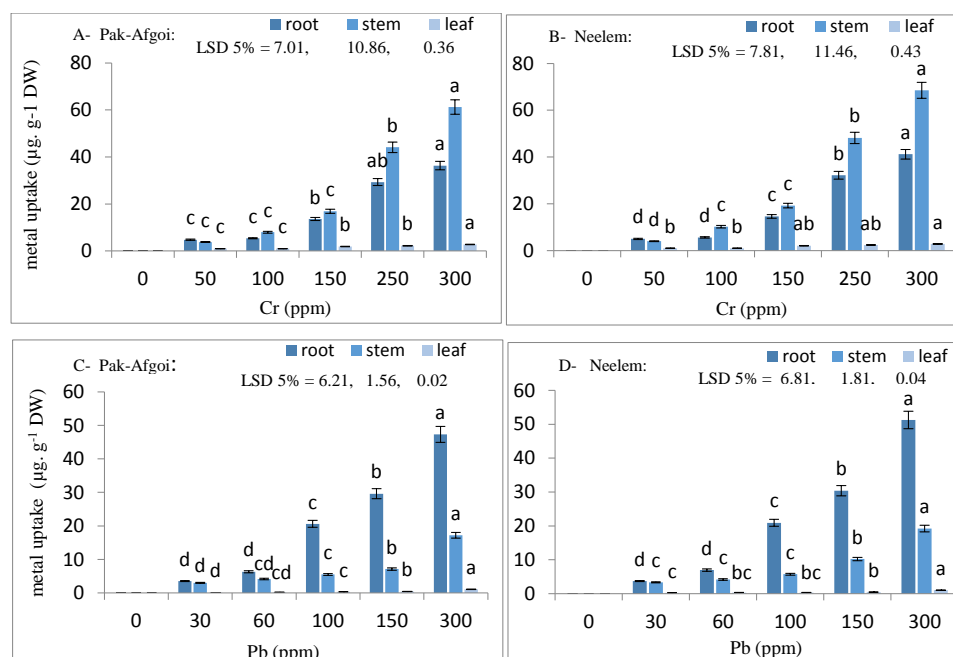


Figure 1 Metal partitioning ($\mu\text{g g}^{-1}$) in different plant parts of Pak-Afgoi & Neelem maize at 40 d (A–D).

Full-size DOI: [10.7717/peerj.16067/fig-1](https://doi.org/10.7717/peerj.16067/fig-1)

to be inhibitory for maize germination and, inhibition exceeded along with the increasing metal concentrations in the medium. Arguments by *Sengar et al. (2008)* revealed inhibition of germination due to interference of Cr and Pb with the essential enzymes for seed germination (amylase and protease). Moreover, *Atici, Agar & Battal (2005)* documented that inhibition of GA_3 (gibberellic acid) and activation of ABA (abscisic acid) during germination of *Cicer arietinum* (chickpea) were due to Zn, Pb, and Cd metals.

Heavy metals are considered the major environmental toxins that adversely affect all living organisms including plants (*Ashraf et al., 2018; Bargagli et al., 2019*). The toxic effects of metals on different growth attributes in plants are due to abnormal nutrient uptake from plant roots as metals become stuck in roots and oppose nutrient uptake from the soil (*Singh et al., 2016*). Different agronomic parameters of rice plants *i.e.*, plant length, tiller count, and dry weight biomass undergoes significant reduction due to Pb doses 0.6 mM–1.2 mM. Observations highlighted less toxicity at a lower Pb dose of 0.6 mM than at 1.2 mM of Pb. A decline in the length of rice plants at the maximum Pb dose was 13% and dry weight decreased by 61% in cultivar Ilmi (*Khan et al., 2021*). A similar observation was reported by *Orhue & Ekhomon (2010)*. Cr affected plant height and dry matter in waterleaf after 100 mg Cr dose. Reduced plant length was due to Cr accumulation that suppressed mitotic activity in the affected plants. Decreases in plant fresh weight due to Cr-toxicity at the varying extent of Cr concentrations also have been reported in *Hibiscus esculentus* (*Amin et al., 2013*), *Helianthus annuus* (*Fozia et al., 2008*), and *Brassica oleracea* (*Ozdener et al., 2011*). These studies also revealed that a decrease in growth and biomass parameters

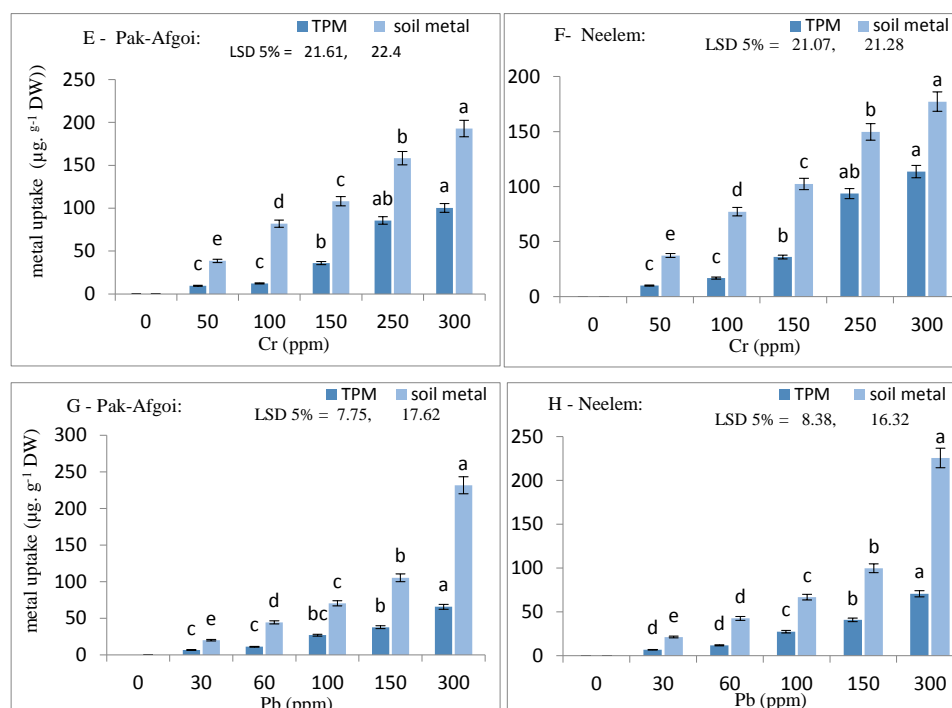


Figure 2 Total metal uptake by plant (TPM) and soil metal retention ($\mu\text{g g}^{-1}$) at 40 d (E–H).

Full-size [DOI: 10.7717/peerj.16067/fig-2](https://doi.org/10.7717/peerj.16067/fig-2)

of the subsequent plants occurs due to increasing metal levels in the growth medium. In agreement with these earlier studies, the present study has also revealed the negative effect of increasing concentrations of Cr and Pb on maize growth (Fig. 4).

Plant growth is attributed mainly to leaf characteristics and the photosynthetic performance of plants. Metals (Cr and Pb) are well-known abiotic stressors that inhibit the photosynthetic performance of intact plants and finally affect plant growth and biomass yield (Houri et al., 2020). Altered values of RGR also predicted the affected plant growth whereas, NAR revealed the negative impact of metals on photosynthetic performance and product of photosynthesis *i.e.*, dry matter content. A leaf is an important photosynthetic organ of plants that plays a key role in the growth of plants. Pb and Cr adversely affect the growth and development of leaves in *Lycopersicon esculantum*, *Pisum sativum*, and *Zea mays* (Yoon et al., 2006; Anjum et al., 2016). Studies showed the inhibitory role of heavy metals on leaf growth and development in rice plants through the generation of oxidative stress/ROS (Singh et al., 2020). These studies are strappingly evident the findings of the present study that leaf number and leaf area in tested plants of maize significantly declined upon exposure to Cr and Pb doses.

Chlorophyll is one of the crucial molecules to facilitate photosynthetic activity in plants and is responsible for the electron transport chain to step forward the photosynthesis. However, heavy metals are responsible for altering the chloroplast structure and cause inhibition of the electron transport system by affecting its biosynthesis (Wakeel, Xu & Gan, 2020) through increased activity of chlorophyllase reported reduced chlorophyll

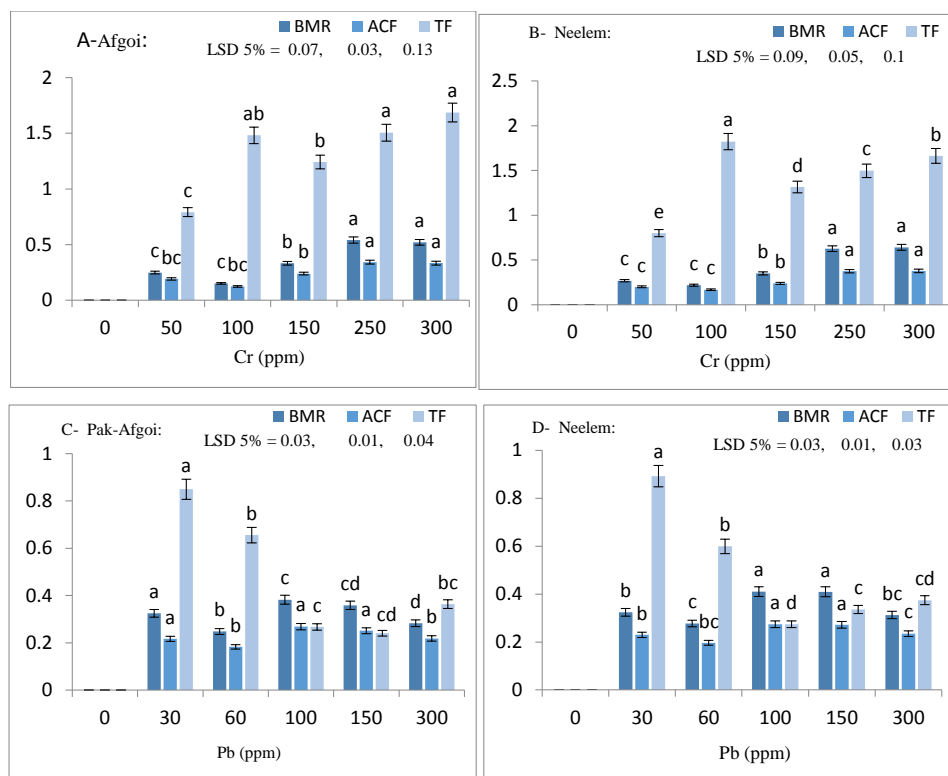


Figure 3 Metal bio-magnification ratio (BMR), metal accumulation factor (ACF) & translocation factor (TF) at day 40 in Pak-Afgoi & Neelem maize (A–D).

Full-size [DOI: 10.7717/peerj.16067/fig-3](https://doi.org/10.7717/peerj.16067/fig-3)

biosynthesis and affected activity of NADPH protochlorophyllide oxidoreductase enzymes under Hg-toxicity. In continuation, *Singh et al. (2020)* investigated Cr⁺⁶-induced alterations up to 89% in the chlorophyll content of mung beans (*Vigna radiata* L). A greater decline was at the highest concentration of 120 μ M than at lower doses of 60–90 μ M. Studies on Cr and Pb stress in *Nicotiana tabacum* and *Cicer arietinum* by *Bukhari et al. (2016)* and *Singh et al. (2020)* have also supported the findings of the present study. Heavy metals cause land degradation through soil acidification that happens due to the leaching mechanism of toxic metal ions. Soils presenting a low pH profile make metals to be available for growing plants and thus reduce crop yield (*Xu et al., 2012*). Although soils pose resistance to the pH change and act as a buffer (*Curtin & Trollove, 2013*); the long-term application of heavy metals put acidification effects on the subsequent soils. Heavy metals undergo hydrolysis in a solution of such soils, generate H⁺ ions, and lower the pH (*Motuzova, Makarichev & Petrov, 2011*; *Schwertfeger & Hendershot, 2012*). Consequently, soil acidification results in nutrient depletion and affects crop plants (*Najafi & Jalali, 2016*).

Soil organic matter (SOM) is assumed as a potential source for microbial activity in the agricultural soils and releases nutrients into the soils through the degradation of soil organic components. Likewise, SOM also shows a large sorption affinity toward metals (*Yin et al., 2002*). Likewise, microbes release certain enzymes of key value into the soil.



Figure 4 Comparison of growth/height of trialed maize plants from two varieties PakAfgoi & Neelem under Cr & Pb treatment.

Full-size  DOI: [10.7717/peerj.16067/fig-4](https://doi.org/10.7717/peerj.16067/fig-4)

These enzymes (urease, dehydrogenase, and phosphatase) are dynamic in the biochemical functionality of the soils including the decomposition of SOM. Hence, the soil enzymes are referred to as an indicator of soil quality, a good measure of soil microbial activity with the recycling of nutrients from the decomposed SOM (Puglisi *et al.*, 2006). Soil urease is a sensitive hydrolyzing enzyme and is a potential indicator of soil pollution and biological activities (Hinojosa, Carreira & García-Ruiz, 2004). Moreover, despite the positive role of urease in soil chemistry; the addition of varying doses of Cu, Cd, Zn, Pb, and Cr decreases the activity of soil urease at large, in contrast depending on the incubation period (Malley, Nair & Ho, 2005; Shen *et al.*, 2005). This might be correlated with the decomposition of SOM.

Soil respiration (CO₂ evolution) is another parameter to assess soil microbial performance with the decomposition of SOM in the subsequent soils (Nawaz *et al.*, 2015). Verma *et al.* (2010) have reported a decreased rate of SOM decomposition under Cd, Cr⁺⁶, and Pb stress vide different incubation periods. Toxicity of all three metals was found to increase towards SOM-decomposition and soil respiration (CO₂ evolution) along with the increasing metal treatment and incubation period. Investigations by Algaïdi (2013)

have extended the Zn, Pb toxicity on aerobic bacteria. An elevated level of both metals significantly decreased the physiological activities of soil microbes and CO₂ production. Similar evidence has been provided by *Mathe-Gaspar et al. (2005)* for Zn, Cu, and Cd metal ions. Soil microbial activity and biodegradation of SOM content play a vital role in the soil fertilization process, cycling of nutrients, and hence increasing the soil fertility (*Kumar et al., 2019*). Likewise, the soil urease enzyme has a potential role in soil N-cycle due to its hydrolytic properties. Consequently, the difference between pre and post-harvest soils revealed a remarkable decline in the available nitrogen under Cr and Pb stress is in agreement with the experimental outcomes of *Orhue & Ekhomun (2010)* i.e., increasing concentration of Cr⁺⁶ continues to decline the soil N by 39% at highest Cr dose 200 mg. A similar observation was reported for soil P and K availability in the present study, indicating the Cd-affected activity of dehydrogenase and phosphatase (*Hassan et al., 2013c*). The present study also revealed similar effects of Cr and Pb on soil macronutrients and indicated the metal toxicity on mineral cycling with the affected SOM and enzymatic activity i.e., urease for N-cycling in the treated soil.

Crops grown on metal-contaminated soils have a greater accumulation of these metals than crops grown in uncontaminated soil (*Sharma, Agrawal & Marshall, 2008*). Plants have a natural capacity to absorb metal ions from the soil even in low concentrations through their root system. To attain efficient reclamation of metal-contaminated soil, plant roots form a rhizosphere ecosystem, absorb and accumulate the heavy metals and improve soil fertility (*Jacob et al., 2018*). Hyper-accumulator plant species were found to be effective in the removal of metals. Plant species that have the potential of accumulating a major portion of metals from the soil are referred to as hyper-accumulators; and are used in phytoremediation techniques to remove the pollutants (*Clemens, 2006*). However, phytoremediation potential exactly depends on the plant's capacity to extract heavy metals from the intact environment and bio-accumulating them in various plant parts without having adverse effects on soil structure, fertility, and biological activity (*Yan et al., 2020*). For instance, *Paspalum fasciculatum* showed the potential of accumulating Cd and Pb in declining order of metal concentrations in roots >leaves >stem. Cd uptake was recorded more than Pb, revealing this plant to be phytostabilizing as the maximum Cd amount was accumulated in roots (*Salas-Moreno & Marrugo-Negrete, 2020*). Likewise, phytostabilization of ryegrass (*Lolium perenne* L.) also showed to be potential for removal of Cd, Cu, As, Zn, and Cr (*Alvarenga et al., 2009*). *Panicum* grass also exhibited a maximum accumulation of Pb in roots than in shoot i.e., roots accumulate 96% more lead as compared to shoot (*Pires-Lira et al., 2020*). In the present study, uptake of Cr and Pb (Afgoi 33%, 22%; Neelem 37%, 24%) at 300 ppm by maize varieties during EGS has uncovered the emergent potential i.e., hyper-accumulation and phytostabilization of this crop cultivated under particular soil and environmental conditions of Dera Ghazi Khan. A higher portion of Cr metal was observed in stem tissues than in roots and leaves. Likewise, Pb accumulation was more prominent in roots than stems and leaves of the subsequent maize plants. The phytoaccumulation potential of maize variety Neelem was more remarkable than Pak-Afgoi, indicating maize varieties as hyper-accumulators for Cr with phytostabilization for Pb metal. These maize plants contain higher Cr and Pb levels than the WHO/FAO

permissible limits *i.e.*, Cr 2.3 mg/kg and Pb: 0.3 mg/kg (Adu, Aderinola & Kusemiju, 2012). The contaminated maize plants are recommended as unsafe for health and be destroyed systematically by burning in high-temperature cement kiln of *D.G. Khan Cement Company* which is available in the study area. Moreover, in the future, screening of different native plant species for phytoremediation purposes along with the focus on their biochemical responses, and tolerance mechanisms is suggested. Application of phosphorus increases soil fertility through increased microbial activity and improves soil nutrient status. The efficacy of soil enzymes (urease, phosphatase) to recycle the nutrients turns high due to phosphorus implication in the contaminated soil (Iqbal *et al.*, 2023). Similarly, the microbial role of bioremediation may be another choice to reclaim contaminated soils. At present, to prevent further addition of heavy metals into agro-soils, irrigation with municipal and industrial wastewater should be banned or if irrigation with the wastewater is continued, it should be recycled through wastewater treatment plants (Atta *et al.*, 2023).

CONCLUSION

Both heavy metals are toxic to seed germination, plant height, leaf development, plant biomass, and chlorophyll content. Moreover, RGR and NAR values of both varieties also indicated the suppressive role of Cr and Pb in the soil medium. The affected plant growth mechanism showed its affinity with the affected soil characteristics under metal stress, which was much more on higher metal levels. Data has revealed an acidic effect of heavy metals on the soil and affected soil respiration by affecting soil microbial activity. Likewise, decreasing levels of soil enzymes have revealed an affected decomposition of SOM content and recycling of soil nutrients. However, besides the toxic effect of metals, maize plants showed great potential in accumulating/partitioning Cr and Pb from the subsequent rhizospheric pot soils. Therefore, it is suggested that maize test varieties (Pak-Afgoi & Neelem Desi) be grown as a tool of phytoremediation in the contaminated agro-soils of Dera Ghazi Khan District. However, such contaminated maize plants are recommended unsafe, and carcinogenic to use due to exceeding amounts of Cr and Pb metals than permissible limits, and be destroyed through cement kiln burning.

ACKNOWLEDGEMENTS

Present work is a part of the Ph. D research of Muhammad Imran Atta. The authors are very grateful to Dr. Muhammad Khuram Afzal, Associate Department of Food Science (Bahauddin Zakariya University, Punjab-Pakistan) for his valuable suggestions and inputs. The authors also acknowledge the significant input and suggestions of Mr. Muhammad Kaleem (Principal Scientific Assistant, PAEC-LAB).

ADDITIONAL INFORMATION AND DECLARATIONS

Funding

This work was supported by Project number (RSP2023R298), King Saud University, Riyadh, Saudi Arabia. The funders had no role in study design, data collection and analysis, decision to publish, or preparation of the manuscript.

Grant Disclosures

The following grant information was disclosed by the authors:
King Saud University, Riyadh, Saudi Arabia: RSP2023R298.

Competing Interests

Habib Ali is an Academic Editor for PeerJ.

Author Contributions

- Muhammad Imran Atta conceived and designed the experiments, performed the experiments, prepared figures and/or tables, and approved the final draft.
- Syeda Sadaf Zehra conceived and designed the experiments, performed the experiments, prepared figures and/or tables, and approved the final draft.
- Habib Ali analyzed the data, prepared figures and/or tables, and approved the final draft.
- Basharat Ali analyzed the data, prepared figures and/or tables, and approved the final draft.
- Syed Naveed Abbas conceived and designed the experiments, authored or reviewed drafts of the article, and approved the final draft.
- Sara Aimen conceived and designed the experiments, authored or reviewed drafts of the article, and approved the final draft.
- Sadia Sarwar conceived and designed the experiments, authored or reviewed drafts of the article, and approved the final draft.
- Ijaz Ahmad conceived and designed the experiments, authored or reviewed drafts of the article, and approved the final draft.
- Mumtaz Hussain conceived and designed the experiments, authored or reviewed drafts of the article, and approved the final draft.
- Ibrahim Al-Ashkar analyzed the data, prepared figures and/or tables, and approved the final draft.
- Dinakaran Elango analyzed the data, prepared figures and/or tables, and approved the final draft.
- Ayman El Sabagh analyzed the data, prepared figures and/or tables, and approved the final draft.

Data Availability

The following information was supplied regarding data availability:

The raw measurements are available in the [Supplementary File](#).

Supplemental Information

Supplemental information for this article can be found online at <http://dx.doi.org/10.7717/peerj.16067#supplemental-information>.

REFERENCES

- Adu AA, Aderinola OJ, Kusemiju V. 2012.** Heavy metals concentration in garden lettuce (*Lactuca sativa* L.) grown along Badagry expressway, Lagos, Nigeria. *Transnational Journal of Science and Technology* **2**:115–130.
- Ahmad MSA, Ashraf M, Tabassam Q, Hussain M, Firdous H. 2011.** Lead (Pb) induced regulation of growth, photosynthesis, and mineral nutrition in maize (*Zea mays* L.) plants at early growth stages. *Biological Trace Element Research* **144**:1229–1239 DOI [10.1007/s12011-011-9099-5](https://doi.org/10.1007/s12011-011-9099-5).
- Aken BV, Correa PA, Schnoor JL. 2009.** Phytoremediation of polychlorinated biphenyls: new trends and promises. *Environmental Science & Technology* **44**:2767–2776 DOI [10.1021/es902514d](https://doi.org/10.1021/es902514d).
- Akhtar S, Iram S. 2017.** In-vitro assessment of heavy metal removal from contaminated agricultural soil by native plant species. *Pakistan Journal of Analytical & Environmental Chemistry* **18**(2):120–128 DOI [10.21743/pjaec/2017.12.12](https://doi.org/10.21743/pjaec/2017.12.12).
- Akinci IE, Akinci S. 2010.** Effect of chromium toxicity on germination and early seedling growth in melon (*Cucumis melo* L.). *African Journal of Biotechnology* **9**(29):4589–4594.
- Algaidi AA. 2013.** Impact of zinc and lead on soil respiration and microbial content under in vitro conditions. *Egyptian Academic Journal of Biological Sciences* **5**(2):45–50 DOI [10.21608/eajbsg.2013.16638](https://doi.org/10.21608/eajbsg.2013.16638).
- Aliu S, Fetahu S, Rozmam L. 2010.** Variation of physiological traits and yield components of some maize hybrids (*Zea mays* L.) in agro-ecological conditions of Kosovo. *Acta Agriculturae Slovenica* **95**:35–41.
- Alvarenga P, Gonçalves A, Fernandes R, De Varennes A, Vallini G, Duarte E, Cunha-Queda A. 2009.** Organic residues as immobilizing agents in aided phytostabilization: (I) effects on soil chemical characteristics. *Chemosphere* **74**:1292–1300 DOI [10.1016/j.chemosphere.2008.11.063](https://doi.org/10.1016/j.chemosphere.2008.11.063).
- Amin H, Arain BA, Amin F, Surhio MA. 2013.** Phytotoxicity of chromium on germination, growth and biochemical attributes of *Hibiscus esculentus* (L.). *American Journal of Plant Sciences* DOI [10.4236/ajps.2013.412302](https://doi.org/10.4236/ajps.2013.412302).
- Anderson JPE. 1982.** Soil respiration. In: Page AL, ed. *Methods of soil analysis. Part 2, chemical and microbiological properties*. 2nd edition. 9. Madison: ASA-SSAAs, 831–871.
- Anderson JM, Ingram JIS. 1993.** *Tropical soil biology and fertility: a handbook of methods*. 2nd edition. Wallingford, UK: C.A.B. International, 221.
- Andra SS, Datta R, Sarkar D, Saminathan SKM, CP Mullens, Bach SBH. 2009.** Analysis of phytochelatin complexes in the lead tolerant vetiver grass *Vetiveria zizanioides*

- (L.) using liquid chromatography and mass spectrometry. *Environmental Pollution* 157:2173–2183 DOI 10.1016/j.envpol.2009.02.014.
- Anjum SA, Ashraf U, Khan I, Tanveer M, Ali M, Hussain I, Wang LC. 2016.** Chromium and aluminum phytotoxicity in maize: morpho-physiological responses and metal uptake. *CLEAN–Soil, Air, Water* 44(8):1075–1084 DOI 10.1002/clean.201500532.
- Ashraf U, Hussain S, Akbar N, Anjum SA, Hassan W, Tang X. 2018.** Water management regimes alter Pb uptake and translocation in fragrant rice. *Ecotoxicology and Environmental Safety* 149:128–134 DOI 10.1016/j.ecoenv.2017.11.033.
- Atici O, Agar G, Battal P. 2005.** Changes in phytohormone contents in chickpea seeds germinating under lead or zinc stress. *Biologia Plantarum* 49(2):215–222 DOI 10.1007/s10535-005-5222-9.
- Atta MI, Zehra SS, Dai DQ, Ali H, Naveed K, Ali I, Sarwar M, Ali B, Bawazeer SI Abdel-Hameed, UK, Ali I. 2023.** Amassing of heavy metals in soils, vegetables and crop plants irrigated with wastewater: Health risk assessment of heavy metals in Dera Ghazi Khan, Punjab, Pakistan. *Frontiers in Plant Science* 13:1080635 DOI 10.3389/fpls.2022.1080635.
- Baker AJM, McGrath SP, Sidoli CMD, Revees RD. 1995.** The possibility of in situ heavy metal decontamination of polluted soils using crops of metal accumulating plants. *Resources, Conservation and Recycling* 11:41–49 DOI 10.1016/0921-3449(94)90077-9.
- Bargagli R, Ancora S, Bianchi N, Rota E. 2019.** Deposition, abatement and environmental fate of pollutants in urban green ecosystems: suggestions from long-term studies in Siena (Central Italy). *Urban Forestry & Urban Greening* 46:126483 DOI 10.1016/j.ufug.2019.126483.
- Bukhari SAH, Wang R, Wang W, Ahmed IM, Zheng W, Cao F. 2016.** Genotype-dependent effect of exogenous 24-epibrassinolide on chromium-induced changes in ultra-structure and physicochemical traits in tobacco seedlings. *Environmental Science and Pollution Research* 23(18):18229–18238 DOI 10.1007/s11356-016-7017-2.
- Causton DR, Venus JC. 1981.** *The biometry of plant growth*. London: Edward Arnold, 307.
- Clemens S. 2006.** Toxic metal accumulation, responses to exposure and mechanisms of tolerance in plants. *Biochimie* 88:1707–1719 DOI 10.1016/j.biochi.2006.07.003.
- Curtin D, Trollove S. 2013.** Predicting pH buffering capacity of New Zealand soils from organic matter content and mineral characteristics. *Soil Research* 51:494–502 DOI 10.1071/SR13137.
- Das R, Jayalekshmy VG. 2015.** Mechanism of heavy metal tolerance and improvement of tolerance in crop plants. *Journal of Global Biosciences* 4(7):2678–2698 DOI 10.1007/s11816-017-0467-2.
- Devi NB, Yadava PS. 2009.** Emission of CO₂ from the soil and immobilization of carbon in microbes in a subtropical mixed oak forest ecosystem, Manipur, Northeast India. *Current Science* 96(12):1627–30.
- Dipu S, Kumar AA, Thanga SG. 2012.** Effect of chelating agents in phytoremediation of heavy metals. *Remediation Journal* 22:133–146 DOI 10.1002/rem.21304.

- Fozia A, Anjum ZA, Ashraf M, Mahmood KZ. 2008.** Effect of Chromium on growth attributes in sunflower (*Helianthus annuus* L.). *Journal of Environmental Sciences*.
- Gardea-Torresdey JL, De la Rosa G, Peralta-Videa JR, Montes M, Cruz-Jimenez G, Cano-Aguilera I. 2005.** Differential uptake and transport of trivalent and hexavalent chromium by tumble weed (*Salsola kali*). *Archives of Environmental Contamination and Toxicology* **48**(2):225–232 DOI [10.1007/s00244-003-0162-x](https://doi.org/10.1007/s00244-003-0162-x).
- Glick BR. 2012.** Plant growth-promoting bacteria: mechanisms and applications. *Scientifica* **2012**:963401 DOI [10.6064/2012/963401](https://doi.org/10.6064/2012/963401).
- Gulser F, Erdogan E. 2008.** The effects of heavy metal pollution on enzyme activities and basal soil respiration of roadside soils. *Environmental Monitoring and Assessment* **145**:127–133 DOI [10.1007/s10661-007-0022-7](https://doi.org/10.1007/s10661-007-0022-7).
- Hasan SA, Fariduddin Q, Ali B, Hayat S, Ahmad A. 2009.** Cadmium: toxicity and tolerance in plants. *Journal of Environmental Biology* **30**(2):165–174.
- Hasanuzzaman M, Bhuyan MHM, Zulfiqar F, Raza A, Mohsin SM, Al-Mahmud J, Fujita M, Vasileios F. 2020.** Reactive oxygen species and antioxidant defense in plants under abiotic stress: revisiting the crucial role of a universal defense regulator. *Antioxidants* **9**(8):681 DOI [10.3390/antiox9080681](https://doi.org/10.3390/antiox9080681).
- Hassan W, Chen W, Ca P, Huang Q. 2013c.** Oxidative enzymes, the ultimate regulator: implications for factors affecting their efficiency. *Journal of Environmental Quality* **42**:1779–1790 DOI [10.2134/jeq2013.05.0204](https://doi.org/10.2134/jeq2013.05.0204).
- Hinojosa MB, Carreira JA, García-Ruiz R. 2004.** Soil moisture pre-treatment effects on enzyme activities as indicators of heavy metal-contaminated and reclaimed soils. *Soil Biology and Biochemistry* **36**:1559–1568 DOI [10.1016/j.soilbio.2004.07.003](https://doi.org/10.1016/j.soilbio.2004.07.003).
- Houri T, Khairallah Y, Al-Zahab A, Osta B, Romanos D, Haddad G. 2020.** Heavy metals accumulation effects on the photosynthetic performance of genotypes in Mediterranean reserve. *Journal of King Saud University - Science* **32**(1):874–880 DOI [10.1016/j.jksus.2019.04.005](https://doi.org/10.1016/j.jksus.2019.04.005).
- IARC. 2012.** Agents classified by the IARC Monographs. Volume 1–123. Available at https://monographs.iarc.fr/wpcontent/uploads/2019/02/List_of_Classification.pdf (accessed on 24 November 2019).
- Iqbal B, Khan I, Javed Q, Fahad-Alabbosh K, Inamullah I, Zhou Z, Rehman A. 2023.** The high phosphorus incorporation promotes the soil enzymatic activity, nutritional status, and biomass of the crop. *Polish Journal of Environmental Studies* **32**(3):158765.
- Islam E, Liu D, Li T, Yang X, Mahmood Q, Tian S, Li J. 2008.** Effect of Pb toxicity on leaf growth, physiology and ultra-structure in the two ecotypes of *Elsholtzia argyi*. *Journal of Hazardous Materials* **154**:914–926 DOI [10.1016/j.jhazmat.2007.10.121](https://doi.org/10.1016/j.jhazmat.2007.10.121).
- Jacob JM, Karthik C, Saratale RG, Kumar SS, Prabakar D, Kadirvelu K. 2018.** Biological approaches to tackle heavy metal pollution: a survey of literature. *Journal of Environmental Management* **217**:56–70 DOI [10.1016/j.jenvman.2018.03.077](https://doi.org/10.1016/j.jenvman.2018.03.077).
- Kandeler E, Gerber H. 1988.** Short-term assay of soil urease activity using colorimetric determination of ammonium. *Biology and Fertility of Soils* **6**:68–72 DOI [10.1007/BF00257924](https://doi.org/10.1007/BF00257924).

- Kanwal A, Farhan M, Sharif F, Hayyat MU, Shahzad L, Ghafoor GZ. 2020.** Effect of industrial wastewater on wheat germination, growth, yield, nutrients and bioaccumulation of lead. *Scientific Reports* **10**:11361 DOI [10.1038/s41598-020-68208-7](https://doi.org/10.1038/s41598-020-68208-7).
- Karkush MO, Ali SD. 2020.** Impacts of lead nitrate contamination on the geotechnical properties of clayey soil. *Journal of Engineering Science and Technology* **15**(2):1032–1045.
- Karkush MO, Zaboony AT, Hussien HM. 2014.** Studying the effects of contamination on the geotechnical properties of clayey soil. In: *Coupled Phenomena in Environmental Geotechnics*. London: Taylor & Francis Group, 599–607.
- Khan M, Khan HN, Aslam H. 2003.** Water quality monitoring of Hudiana Drain. *Pakistan Journal of Biological Sciences* **6**:167–173 DOI [10.3923/pjbs.2003.167.175](https://doi.org/10.3923/pjbs.2003.167.175).
- Khan M, Ibrahim TN, Al Azzawi, Imran M, Hussain A, Mun BG, Pande A, Yun BW. 2021.** Effects of lead (Pb) induced oxidative stress on morphological and physio-biochemical properties of rice. *Agronomy* **11**(3):409 DOI [10.3390/agronomy11030409](https://doi.org/10.3390/agronomy11030409).
- Kopittke PM, Asher CJ, Menzies NW. 2008.** Prediction of Pb speciation in concentrated and dilute nutrient solutions. *Environmental Pollution* **153**(3):548–554 DOI [10.1016/j.envpol.2007.09.012](https://doi.org/10.1016/j.envpol.2007.09.012).
- Kouser A, Khan AA. 2021.** Chromium induced changes in growth and physiological attributes of Chicory (*Cichorium intybus* L.), an important medicinal plant. *Plant Science Today* **8**(3):509–516 DOI [10.14719/pst.2021.8.3.1120](https://doi.org/10.14719/pst.2021.8.3.1120).
- Kumar A, Dewangan S, Lawate P, Bahadur I, Prajapati S. 2019.** Zinc-solubilizing bacteria: a boon for sustainable agriculture, in plant growth promoting rhizobacteria for sustainable stress management. In: *Rhizobacteria in abiotic stress management*. Volume 1. US: Springer, 139–155 DOI [10.1007/978-981-13-6536-2_8](https://doi.org/10.1007/978-981-13-6536-2_8).
- Li G, Zhao X, Iqbal B, Zhao X, Liu J, Javed Q, Du D. 2023.** The effect of soil microplastics on *Oryza sativa* (L.) root growth traits under alien plant invasion. *Frontiers in Ecology and Evolution* **11**:1172093 DOI [10.3389/fevo.2023.1172093](https://doi.org/10.3389/fevo.2023.1172093).
- Ludwig B, Khanna PK, Anuragsa B, Folster H. 2001.** Assessment of cation and anion exchange and pH buffering in an Amazonian Ultisol. *Geoderma* **102**:27–40 DOI [10.1016/S0016-7061\(00\)00099-9](https://doi.org/10.1016/S0016-7061(00)00099-9).
- Malley C, Nair J, Ho G. 2005.** Impact of heavy metals on enzymatic activity of substrate and on composting worms *Eisenia fetida*. *Bioresource Technology* **97**:1498–1502 DOI [10.1016/j.biortech.2005.06.012](https://doi.org/10.1016/j.biortech.2005.06.012).
- Marques APGC, Rangel AOSS, Castro PML. 2009.** Remediation of heavy metal contaminated soils: phytoremediation as a potentially promising clean-up technology. *Critical Reviews in Environmental Science and Technology* **39**:622–654 DOI [10.1080/10643380701798272](https://doi.org/10.1080/10643380701798272).
- Mathe-Gaspar G, Mathe PL, Szabo B, Uzinger N, Anton A. 2005.** After-effect of heavy metal pollution in brown forest soils. Proceedings of the 8th Hungarian congress on plant physiology and the 6th Hungarian conference on photosynthesis. *Acta Biologica Szegediensis* **49**:71–72 DOI [10.5402/2011/402647](https://doi.org/10.5402/2011/402647).

- Moreno JL, Garcia C, Hernandez T. 2003.** Toxic effect of cadmium and nickel on soil enzymes and the influence of adding sewage sludge. *European Journal of Soil Science* 54:377–386 DOI [10.1046/j.1365-2389.2003.00533.x](https://doi.org/10.1046/j.1365-2389.2003.00533.x).
- Motuzova GV, Makarichev IP, Petrov MI. 2011.** The effects of iron, mercury, and copper ions on the acid–base properties of aqueous soil extracts. *Moscow University Soil Science Bulletin* 66:157–162 DOI [10.3103/S0147687411040041](https://doi.org/10.3103/S0147687411040041).
- Murakami M, Ae N, Ishikawa S. 2007.** Phytoextraction of cadmium by rice (*Oryza sativa* L), soybean (*Glycine max* L), and maize (*Zea mays* L). *Environmental Pollution* 145:96–103 DOI [10.1016/j.envpol.2006.03.038](https://doi.org/10.1016/j.envpol.2006.03.038).
- Mussarat M, Bhatti AU, Khan FU. 2007.** Concentration of metals in sewage and canal water used for irrigation in Peshawar. *Sarhad Journal of Agriculture* 23:335–338.
- Najafi S, Jalali M. 2016.** Effect of heavy metals on pH buffering capacity and solubility of Ca, Mg, K, and P in non-spiked and heavy metal-spiked soils. *Environmental Monitoring and Assessment* 188:342 DOI [10.1007/s10661-016-5329-9](https://doi.org/10.1007/s10661-016-5329-9).
- Naseem S, Yasin M, Ahmed A, Faisal M. 2015.** Chromium accumulation and toxicity in corn (*Zea mays* L.) seedlings. *Polish Journal of Environmental Studies* 24(2):899–904.
- Nawaz M, Wahid A, Ahmad SS, Butt A. 2015.** Response of soil microbial biomass and respiration in heavy metal contaminated soil of Multan. *International Journal of Biosciences* 7(4):68–77 DOI [10.12692/ijb/7.4.68-77](https://doi.org/10.12692/ijb/7.4.68-77).
- Olsen SR, Sommers LE. 1982.** Phosphorus. In: Page AL, Miller RH, eds. *Methods of soil analysis. Part 2. 2nd (Eds) Agronomy monograph 9*. Madison: ASA and SSSA, 403–430.
- Orhue ER, Ekhmun AM. 2010.** Chromium effects on growth of early water leaf (*Talinum triangulare*) in an Ulitisol. *American-Eurasian Journal of Agricultural and Environmental Sciences* 7(5):586–590.
- Ozdener Y, Aydin B, Fatma-Aygün S, Yürekli F. 2011.** Effect of hexavalent chromium on the growth and physiological and biochemical parameters on *Brassica oleracea* L. var. acephala DC. *Acta Biologica Hungarica* 62(4):463–476 DOI [10.1556/ABiol.62.2011.4.11](https://doi.org/10.1556/ABiol.62.2011.4.11).
- Peternella WS, da Silva FF, da Costa ACS. 2021.** Evaluation of the phytoavailability of Cu (II) and Cr (III) for the growing of Corn (*Zea mays* L.), cultivated in four soils of a topo-sequence derived from Basalt. *Open Access Library Journal* 8(8):1–21 DOI [10.4236/oalib.1107707](https://doi.org/10.4236/oalib.1107707).
- Pires-Lira MF, de Castro EM, Lira JMS, de Oliveira C, Pereira FJ, Pereira MP. 2020.** Potential of *Panicum aquaticum* Poir, for the phytoremediation of aquatic environments contaminated by lead. *Ecotoxicology and Environmental Safety* 193:110336 DOI [10.1016/j.ecoenv.2020.110336](https://doi.org/10.1016/j.ecoenv.2020.110336).
- Puglisi E, Del Rea AAM, Robab MA, Gianfredab L. 2006.** Development and validation of numerical indexes integrating enzyme activities of soils. *Soil Biology and Biochemistry* 38:1673–1681 DOI [10.1016/j.soilbio.2005.11.021](https://doi.org/10.1016/j.soilbio.2005.11.021).
- Rafique U, Nasreen S, Tufai FL, Ashra MA. 2016.** Remediation of deltamethrin contaminated cotton fields: residual and adsorption assessment. *Open Life Sciences* 11:417–426 DOI [10.1515/biol-2016-0055](https://doi.org/10.1515/biol-2016-0055).

- Reeves RD, Baker AJ, Jaffré T, Erskine PD, Echevarria G, van Der Ent A. 2018. A global database for plants that hyperaccumulate metal and metalloid trace elements. *New Phytologist* 218(2):407–411 DOI 10.1111/nph.14907.
- Salas-Moreno M, Marrugo-Negrete J. 2020. Phytoremediation potential of Cd and Pb-contaminated soils by *Paspalum fasciculatum* Willd, ex Flüggé. *International Journal of Phytoremediation* 22:87–97 DOI 10.1080/15226514.2019.1644291.
- Schwertfeger DM, Hendershot WH. 2012. Comparing soils chemistries of leached and non-leached copper amended soils. *Environmental Toxicology and Chemistry* 31:2253–2260 DOI 10.1002/etc.1904.
- Sengar RS, Gautam M, Garg SK, Chaudhary R, Sengar K. 2008. Effect of lead on Seed germination, seedling growth, chlorophyll content and nitrate reductase activity in mung bean (*Vigna radiate* L.). *Journal of Photochemistry and Photobiology A* 2(2):61–68 DOI 10.3923/rjphyto.2008.61.68.
- Sharma RK, Agrawal M, Marshall FM. 2008. Heavy metal (Cu, Zn, Cd and Pb) contamination of vegetables in urban India: a case study in Varanasi. *Environmental Pollution* 154:254–263 DOI 10.1016/j.envpol.2007.10.010.
- Sheldrick BH, Wang C. 1993. Particle size distribution. In: Carter MR, ed. *Soil sampling and methods of analysis*. Boca Raton: Canadian Society of Soil Science, Lewis Publishers, 499–512.
- Shen G, Lu Y, Zhou Q, Hong J. 2005. Interaction of polycyclic aromatic hydrocarbons and heavy metals on soil enzyme. *Chemosphere* 61:1175–1182 DOI 10.1016/j.chemosphere.2005.02.074.
- Singh D, Sharma NL, Singh CK, Sarkar SK, Singh I, Dotaniya ML. 2020. Effect of chromium (VI) toxicity on morpho-physiological characteristics, yield, and yield components of two chickpea (*Cicer arietinum* L.) varieties. *PLOS ONE* 15:e0243032 DOI 10.1371/journal.pone.020243032.
- Singh S, Parihar P, Singh R, Singh VP, Parsad SM. 2016. Heavy metal tolerance in plants: role of transcriptomics, metabolomics, and ionomics. *Frontiers in Plant Science* 6:1143 DOI 10.3389/fpls.2015.01143.
- Uddin MK. 2016. A review on the adsorption of heavy metals by clay minerals, with special focus on the past decade. *Chemical Engineering Journal* 308:438–462 DOI 10.1016/j.cej.2016.09.029.
- Usman K, Abu Dieyeh MH, Zouari N, Ghouti MAI. 2020. Lead (Pb) bioaccumulation and antioxidative responses in *Tetraena qataranse*. *Scientific Reports* 10(1):17070 DOI 10.1038/s41598-020-73621-z.
- Verma RK, Yadav DV, Singh CP, Suman A, Gaur A. 2010. Effect of heavy metals on soil respiration during decomposition of sugarcane (*Saccharum officinarum* L.) trash in different soils. *Plant, Soil and Environment* 56(2):76–81 DOI 10.17221/1773-PSE.
- Wakeel A, Xu M, Gan Y. 2020. Chromium-induced reactive oxygen species accumulation by altering the enzymatic antioxidant system and associated cytotoxic, genotoxic, ultrastructural, and photosynthetic changes in plants. *International Journal of Molecular Sciences* 21:728 DOI 10.3390/ijms21030728.

- Walkley A, Black CA. 1934.** An examination of the Degtjareff method for determining soil organic matter and a proposed modification of the chromic acid titration method. *Soil Science* 37:29–38 DOI 10.1097/00010694-193401000-00003.
- Wang W, Vinocur B, Altman S. 2003.** Plant responses to drought, salinity and extreme temperatures: towards genetic engineering for stress tolerance. *Planta* 218:1–14 DOI 10.1007/s00425-003-1105-5.
- Wang BO, Yanli D. 2013.** Cadmium and its neurotoxic effects-review article. *Oxidative Medicine and Cellular Longevity* 2013:898034 DOI 10.1155/2013/898034.
- Wardle DA, Ghani A. 1995.** A critique of the microbial metabolic quotient (qCO₂) as a bioindicator of disturbance and ecosystem development. *Soil Biology and Biochemistry* 27:1601–1610 DOI 10.1016/0038-0717(95)00093-T.
- Welz B, Sperling M. 1999.** *Atomic absorption spectrometry*. 3rd. Weinheim: Wiley-VCH Google Scholar.
- World Health Organization. 2010.** Action is needed on chemicals of major public concern. *Public Health Environ.* 1–4. Available at https://www.who.int/ipcs/assessment/public_health/chemicals_phc/en/ (accessed on 27 November 2019).
- Wuana RA, Okieimen FE. 2011.** Heavy metals in contaminated soils: a review of sources, chemistry, risks and best available strategies for remediation. *ISRN Ecology* 2011:402647 DOI 10.5402/2011/402647.
- Xu RK, Zhao AZ, Yuan JH, Jiang J. 2012.** pH buffering capacity of acid soils from tropical and subtropical regions of China as influenced by incorporation of crop straw biochars. *Journal of Soils and Sediments* 12:494–502 DOI 10.1007/s11368-012-0483-3.
- Yan A, Tan SN, Yousf MLM, Gosh S, Chen Z. 2020.** Phytoremediation: a promising approach for revegetation of heavy metal-polluted land. *Frontiers in Plant Science* 11:359 DOI 10.3389/fpls.2020.00359.
- Yanqun Z, Yuan L, Jianjun C, Haiyan C, Li Q, Schwartz C. 2005.** Hyperaccumulation of Pb, Zn and Cd in herbaceous grown on lead–zinc mining area in Yunnan, China. *Environment International* 31(5):755–762 DOI 10.1016/j.envint.2005.02.004.
- Yin Y, Impellitteri CA, You SJ, Allen HE. 2002.** The importance of organic matter distribution and exact soil: solution ratio on the desorption of heavy metals from soils. *Science of The Total Environment* 287:107–119 DOI 10.1016/S0048-9697(01)01000-2.
- Yoon J, Cao X, Zhou Q, Ma LQ. 2006.** Accumulation of Pb, Cu, and Zn in native plants growing on a contaminated Florida site. *Science of The Total Environment* 368:456–464 DOI 10.1016/j.scitotenv.2006.01.016.
- Zhang XL, Yan J, Liu Z, Zhang Tan C. 2019.** Removal of different kinds of heavy metals by novel PPG-nZVI beads and their application in simulated storm water infiltration facility. *Applied Sciences* 9:4213 DOI 10.3390/app9204213.